# Atomistic simulations of nanoparticle generation by short pulse laser ablation of AgCu bilayers in liquid

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Interested to learn more? Do not hesitate to contact me!

# Effect of liquid confinement on surface modification



surface micro/nano-structuring by laser processing in liquids

mechanical confinement + additional cooling channel  $\rightarrow$  nonequilibrium microstructure & smoothing of surface morphology



Surface features produced by single shot ablation of Cr (001) target irradiated at 6000 J/m<sup>2</sup> incident laser fluence in air and water Shih, Gnilitskyi, Shugaev, Skoulas, Stratakis, and Zhigilei, *Nanoscale* **12**, 7674, 2020

### Type of the liquid plays a role

5 ps laser processing of Ti in ethanol and water

Barmina *et al.*, *Quant. Electron.* **40**, 1012, 2010



# Laser ablation in liquids for synthesis of nanoparticles

#### synthesis of clean colloidal nanoparticles with unique shapes and functionalities



Wagener et al., Photonik Int. 20, 2011

Sylvestre et al., Appl. Phys. A 80, 758, 2005

clean NPs for biomedicine, catalysis, plasmonics, *etc*.

size, structure and composition of NPs can be controlled by *T*, viscosity of liquid medium, surfactants, PLFL, *etc*.



Marzun et al., Appl. Surf. Sci. 348, 75, 2015

#### TTM-MD model for laser interaction with metals in liquid environment



**TTM-MD model:** MD is combined with TTM to account for (1) laser energy absorption by conduction band electrons, (2) electron-phonon equilibration, (3) electronic heat conduction

**Coarse-grained model for liquids:** heat bath approach accounts for missing degrees of freedom **Acoustic impedance matching boundary conditions:** nonreflective propagation of stress waves

#### "mosaic" approach to mapping processes occurring at the scale of the whole laser spot



Wu and Zhigilei, *Appl. Phys. A* **114**, 11, 2014. Shugaev *et al.*, *MRS Bull.* **41**, 960, 2016.

# Summary on short (fs/ps) pulse laser ablation in liquids



Two mechanisms of NP generation in fs/ps PLAL:

- 1. Rapid nucleation & growth in water-metal mixing region  $\rightarrow$  small ( $\leq$  10 nm) NPs
- 2. Rayleigh-Taylor and Richtmyer-Meshkov instabilities at interface between superheated metal layer and water  $\rightarrow$  large (10s of nm) NPs

Rapid quenching  $(10^{12} \text{ K/s}) \rightarrow \text{NPs}$  with complex microstructure and, possibly, nonequilibrium/metastable phases/structures





Shih et al., J. Phys. Chem. C 121, 16549, 2017
Shih et al., J. Colloid Interface Sci. 489, 3, 2017
Shugaev et al., Appl. Surf. Sci. 417, 54, 2017
Shih et al., Nanoscale 10, 6900, 2018

## The effect of pulse duration on nanoparticle generation in PLAL



Shih et al., Phys. Chem. Chem. Phys. 22, 7077, 2020

# **Spatially modulated ablation in water**



# **Effect of water environment on surface morphology**

TEM images of ripple cross-sections and corresponding simulated surface structures







in water

Shih et al., Nanoscale 12, 7674, 2020

### Laser ablation of AgCu bilayer thin films in water



Shih et al., J. Phys. Chem. C 125, 2132, 2021

#### Laser ablation of AgCu bilayer thin films in water







- Rapid deceleration of the ablation plume by water;
- Accumulation of the plume at the interface with water;
- Slower steady upward movement of the interface with a velocity of  $\sim$ 35 m/s.
- water is blanked to expose the processes occurring in ablation plume



Stratification of the interfacial region into three parts:

- 1. top region of rapid nucleation and growth of numerous small nanoparticles,
- 2. complex coarse morphology of interconnected liquid regions,
- 3. continuous thin metal layer that retains its integrity up to  $\sim 2500$  ps.



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Shih et al., J. Phys. Chem. C 125, 2132, 2021



Same qualitative picture, but some quantitative differences related to  $G_{Cu} > G_{Au}$ :

- 1. less vigorous initial expansion,
- 2. lower *T* of the vapor in the bottom part of the system,
- 3. more numerous large (10s nm) droplets in the upper part of the interfacial region and smaller largest droplet at the lower part of the interfacial region.



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- Limited mixing prior to the formation of the nanoparticles.
- Central parts of the distributions (correspond to well-mixed compositions) are depleted of nanoparticles.
- Limited mixing is surprising, since fluence is  $\sim 3$  times the ablation threshold

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Glass/Cu/Ag 39 nm Ag + 40 nm Cu

Glass/Ag/Cu 65 nm Cu + 23 nm Ag

# **Experimental verification**











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## **Experimental verification**



Shih et al., J. Phys. Chem. C 125, 2132, 2021

#### **Nanoparticles larger than film thickness?**



Diameters of the largest nanoparticles > thickness of the original bilayer films.





4,813,971 atoms,  $D_e = 52.0$  nm, C = 68 at.% Ag



Nanoparticles larger than film thickness?

Accumulation of plume at the interface with water

Formation and decomposition of a transient liquid layer

# $\hat{\Gamma}$

Nanoparticles more than twice larger than the thickness of the original bilayer films

3,969,848 atoms,  $D_e = 45.9$  nm, C = 20 at.% Ag

#### **Summary**

Two surprising observations:

**1.** Nanoscale spatial separation of the two components in the bilayer leads to a sharp departure from the complete quantitative mixing in the colloidal nanoparticles.

**2.** The largest nanoparticles can exceed the thickness of the film.



Explained by complex dynamic interaction between the ablation plume and liquid environment: Accumulation of plume at the interface with liquid  $\rightarrow$  formation and decomposition of a hot liquid layer that prevents mixing and yields large nanoparticles

# **Coarse-grained MD representation of liquid environment**



Heat bath approach accounts for missing internal degrees of freedom

Tabetah et al., J. Phys. Chem. B 118, 13290, 2014; Shih et al., J. Colloid Interface Sci. 489, 3, 2017

properties of water	experiment	CG model	$\Delta$ , %
density, $\rho$ , g/cm <sup>3</sup>	1.0	1.0	0
heat capacity, $c_p$ , J/(kg K)	$4.2 \times 10^{3}$	$4.2 \times 10^{3}$	0
bulk modulus, <i>K</i> , GPa	2.2	1.8	18
speed of sound, $c_s$ , m/s	1483	1342	9
melting temperature, $T_m$ , K	273	330	21
critical temperature, $T_c$ , K	647	520	20
critical density, $\rho_c$ , g/cm <sup>3</sup>	0.322	0.398	24
viscosity, $\eta$ , cP	0.894	0.910	2
surface energy, $\sigma$ , J/m <sup>2</sup>	0.072	0.073	1

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http://www.slims.polimi.it/





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at the 10<sup>th</sup> International Conference on Multiscale Materials Modeling, Baltimore, Maryland (November 7-11, 2021)

https://mmm10.jhu.edu/symposia/Computer-modeling-of-laser-and.html