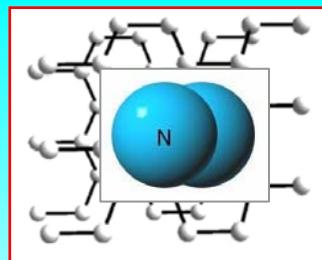


Sarov, 2011

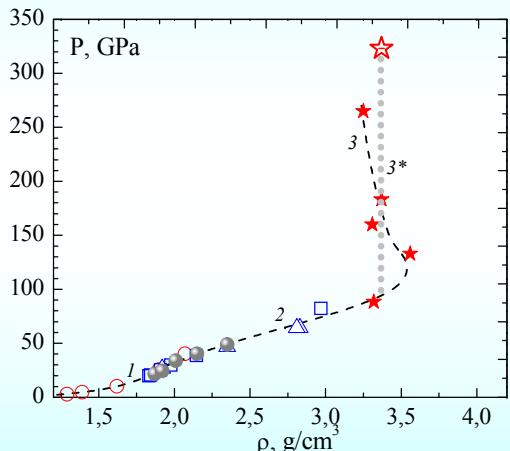
Annual Moscow Workshop
Non-ideal Plasma Physics
Russian Academy of Science
November 23-24, 2011



Chernogolovka, 2010
EMMI_GSI, 2011

Shock compression of liquid nitrogen

Thermodynamic analysis *of new experimental data in megabar range*



Igor Iosilevskiy

*Joint Institute for High Temperature (Russian Academy of Science)
Moscow Institute of Physics and Technology (State University)*

Victor Gryaznov

Institute of Problem of Chemical Physics (Russian Academy of Science)



Shock compression of liquid nitrogen

(new experimental data in megabar pressure range)

Mochalov M., Zhernokletov M., Il'kaev R., Mikhailov A., Mezhevov A., Kovalev A.,
Kirshanov S., Grigorieva Yu., Novikov M., Shuikin A.,
Fortov V., Gryaznov V., Iosilevskiy I.

1

JETP, 137, 77 (2010)

*Density, temperature and electroconductivity measurements in shock compressed
condensed nitrogen at megabar pressure*

2

Trunin R., Boriskov G., Bykov A., Medvedev A., Simakov G., Shuikin A.

JETP Letters, 88 (3), 189 (2008)

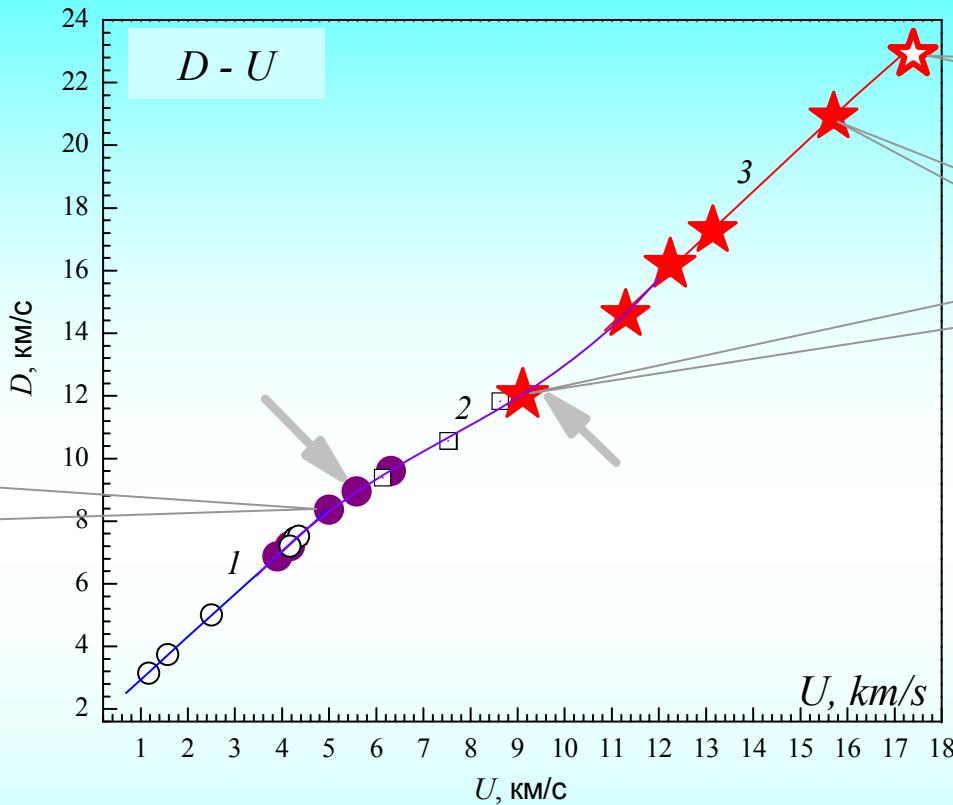
Shock compression of condensed nitrogen at megabar pressure



Collaboration:
Eugene & Lidia Yakyb
(Odessa University, Ukraine)

Shock Hugoniot of liquid nitrogen

(primary experimental data in kinematic variables)



Mochalov et al.
2010
(flat geometry)

Trunin et al.
(2008)

Mochalov et al.
2010
(spherical)

Initial state
 $\rho_0 = 0.81$
 $(T_0 = 77 \text{ K})$

D and U – shock and mass velocities

Arrows – hypothetical transitions between molecular (1),
polymeric (2) and plasma (3) states

D	U
12.03 ± 0.25	9.11
14.60 ± 0.26	11.29
16.19 ± 0.36	12.24
17.28 ± 0.40	13.14
20.90 ± 0.68	15.71
22.9 ± 0.4	17.4

Approximations (Mochalov et al. 2010):

- 1: $D = 1.365 \cdot U + 1.572$
- 2: $D = 3.375 \cdot U - 0.337 \cdot U^2 + 0.015 \cdot U^3 - 2.008$
- 3: $D = 1.407 \cdot U - 1.174$

★ - Mochalov M., Zhernokletov M. et al.,
JETF **137**, 77 (2010)

★ - Trunin R., Boriskov G., Bykov A., Medvedev A.
JETF Letters, **88**, 220 (2008)

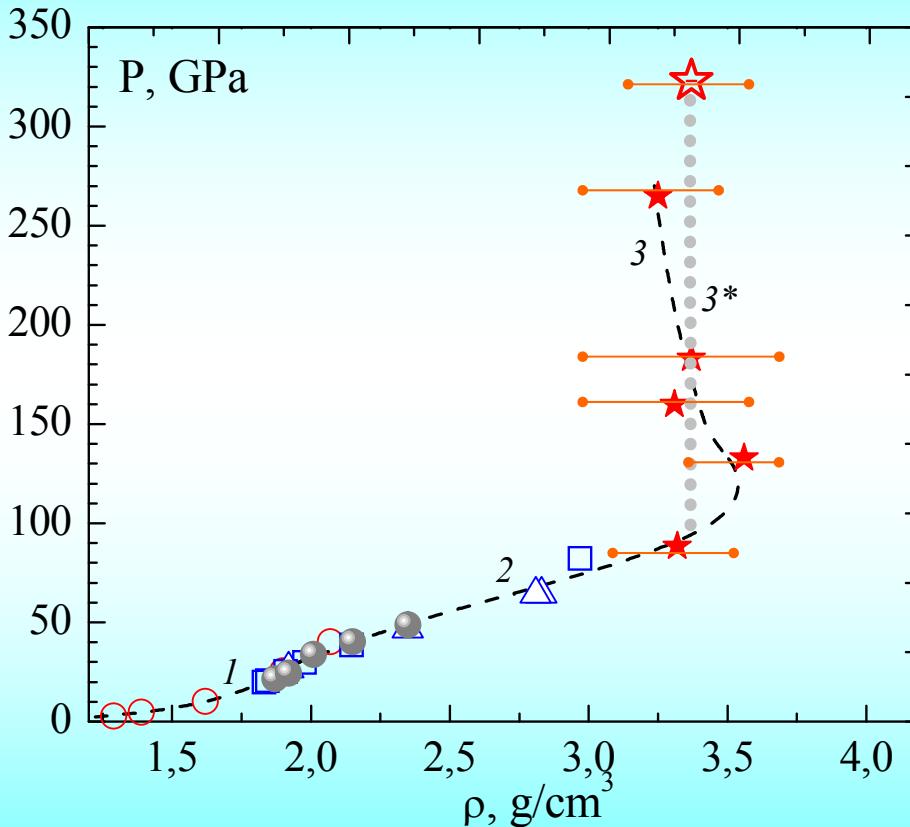
Shock Hugoniot of liquid nitrogen

(experimental data in thermodynamic variables)

$$p_1 = p_0 + \rho_0 D U$$

$$h_1 = h_0 + \left(\frac{D^2}{2} - \frac{(D-U)^2}{2} \right)$$

$$\rho_1 = \rho_0 \{D/(D-U)\}$$



(*) $(\rho/\rho_0)_{\text{ideal gas}} = 4.0$

★ - Mochalov M., Zhernokletov M. et al., *JETP* **137**, 77 (2010)

★ - Trunin R., Boriskov G., Bykov A., Medvedev A. *JETP Letters*, **88**, 220 (2008)

ρ/ρ_0
4.12
4.41
4.10
4.18
4.03
4.16

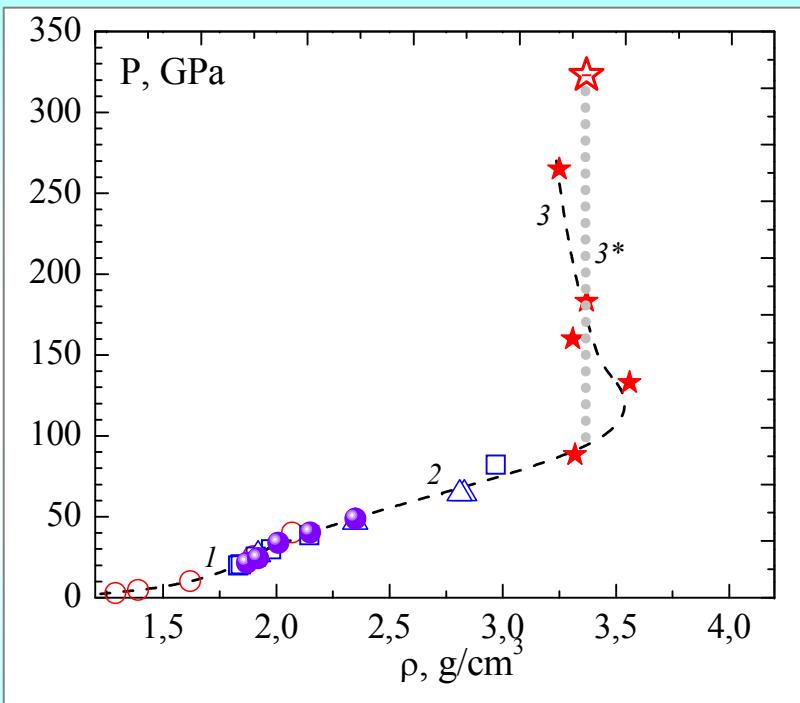
Internal energy is close to be linear function of pressure!

$$E_1 = E_0 + \frac{1}{2} \left(\frac{1}{\rho_0} - \frac{1}{\rho_1} \right) (p_1 + p_0)$$

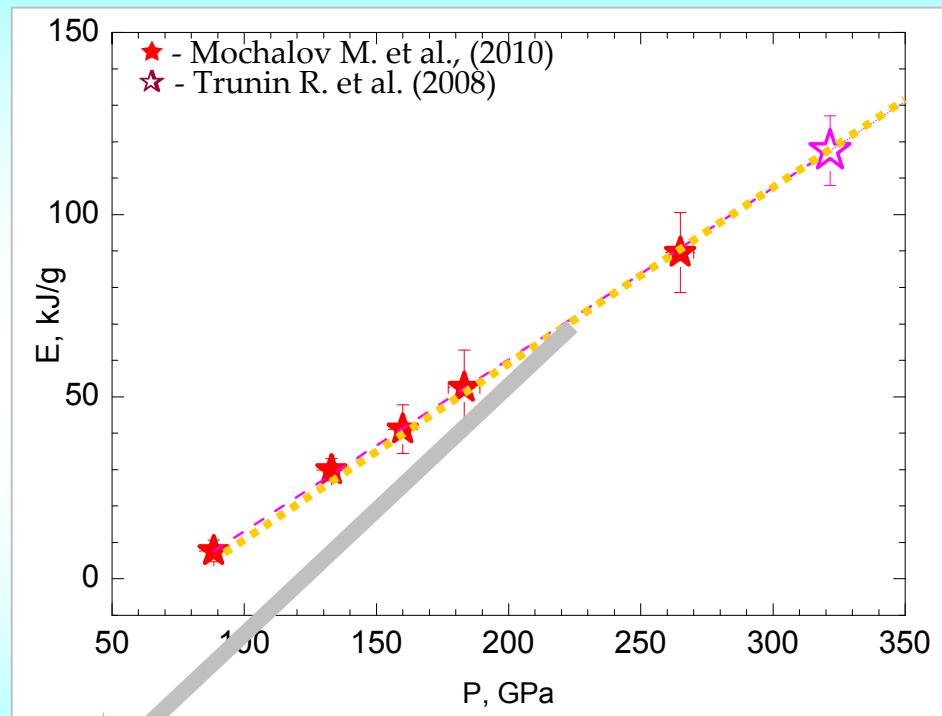


Approximately linear function: $E \sim E_0 + \text{const} \cdot P$

Hugoniot: $\sim \rho \approx \text{const}$ (3.3 ± 0.1 g/cc)



Internal Energy \Leftrightarrow Pressure of shock compressed nitrogen



NB !

$$\text{Gr} = V(\partial P / \partial E)_V \approx \text{const} \approx 0.62$$

(*) $E(P, T)$ reference point – energy of ideal atomic gas N at $T = 0$ K

$$(*) \text{Gr}_{\text{id.gas}} = 2/3 \approx 0.67$$

What does it mean ?

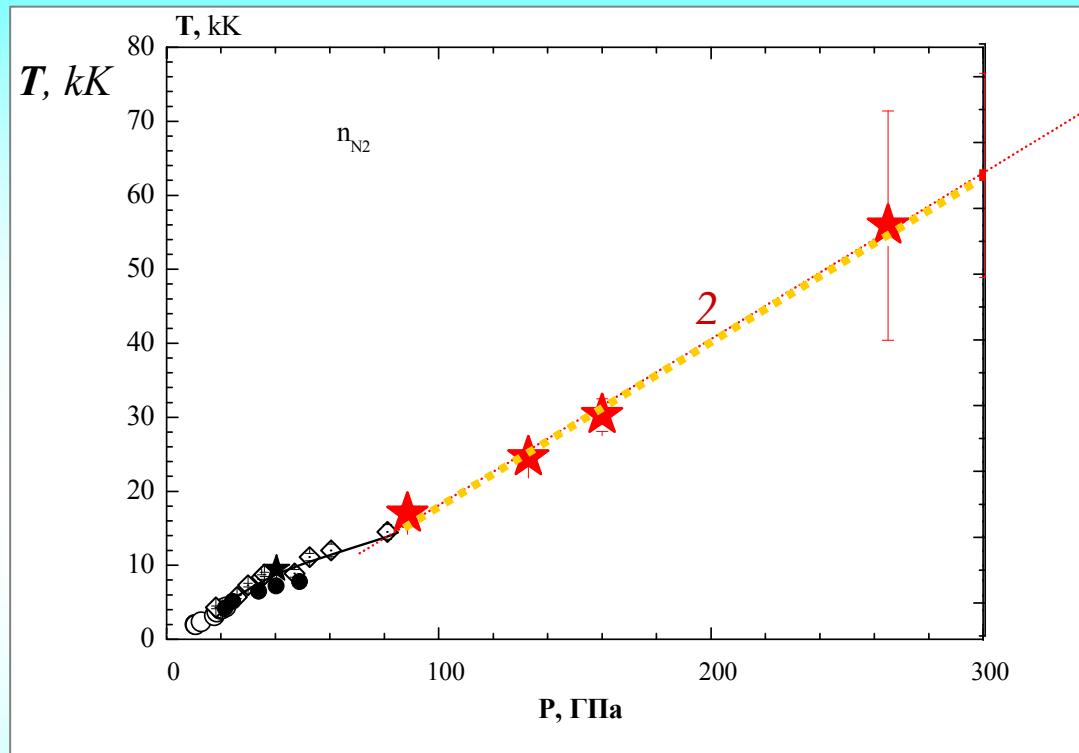
The region of approximate constancy for Gruneisen parameter $\Leftrightarrow \rho \approx 3.3 \pm 0.1$ g/cc $90 < P < 330$ GPa

Temperature of shock compressed nitrogen

measurements of thermal equation of state

P , GPa	T , kK
88.4 ± 2	16.2 ± 0.9
133.3 ± 3	24.6 ± 0.5
160.0 ± 3	28.4 ± 2.2
183.0 ± 3	-
265.0 ± 5	56.0 ± 15.2
323	-

JETF 137, 77 (2010)



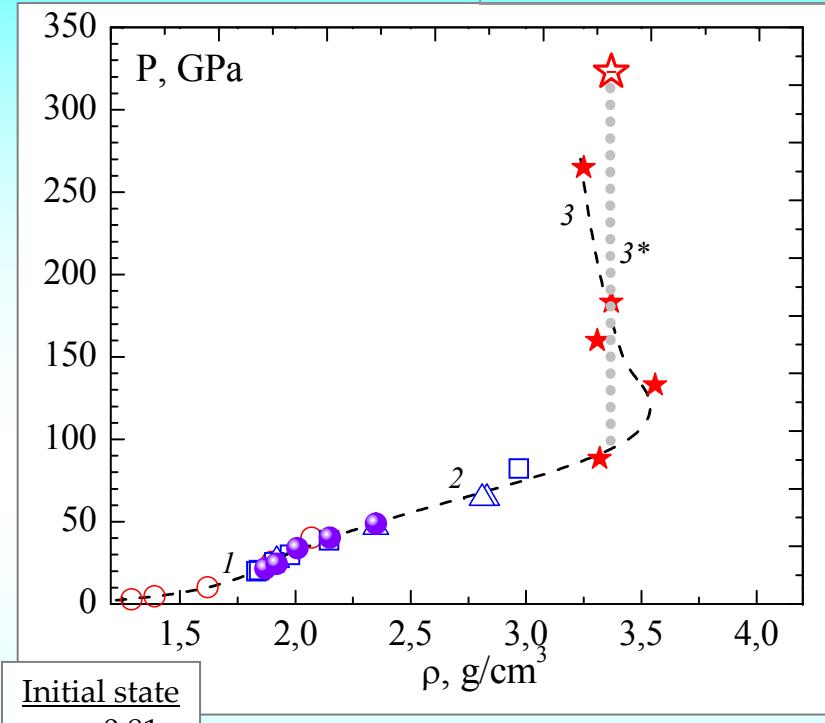
★ - Experiment (Mochalov M., Zhernokletov M. et al., JETF 137, 77 (2010))

2 – linear approximation of experimental data ($P \approx 90 - 300$ ГПа)

Temperature is close to be linear function of pressure + isochoric behavior of nitrogen Hugoniot

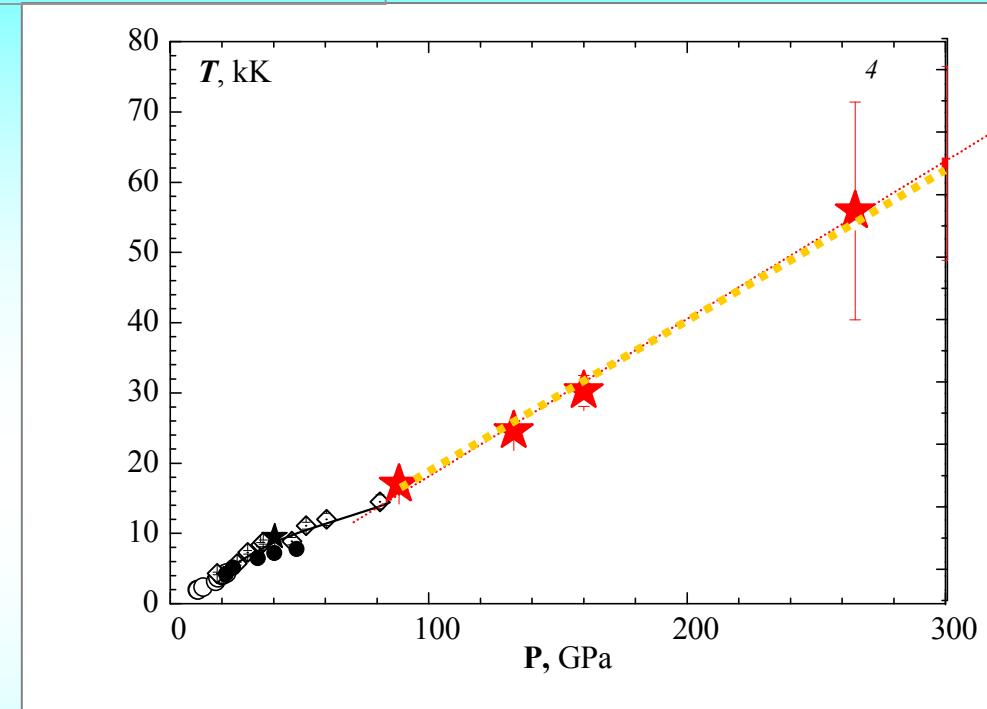
$$\rho \approx 3.3 \pm 0.1 \text{ g/cc}$$

$$(90 < P < 330 \text{ GPa})$$



Initial state

$$\rho_0 = 0.81$$



★ - Mochalov et al., JETP (2010)

Compressibility factor (PV/RT) is almost constant on nitrogen isochore

NB !

$$Z \equiv PV/RT \equiv P/n_N kT \approx \text{const} \approx 2.66 \pm 0.20$$

What does it mean ?

$$\frac{PV}{RT}$$

Ideal gas N₂: Z= 0.5

Ideal gas N: Z= 1.0

Id. gas N⁺ + e: Z= 2.0

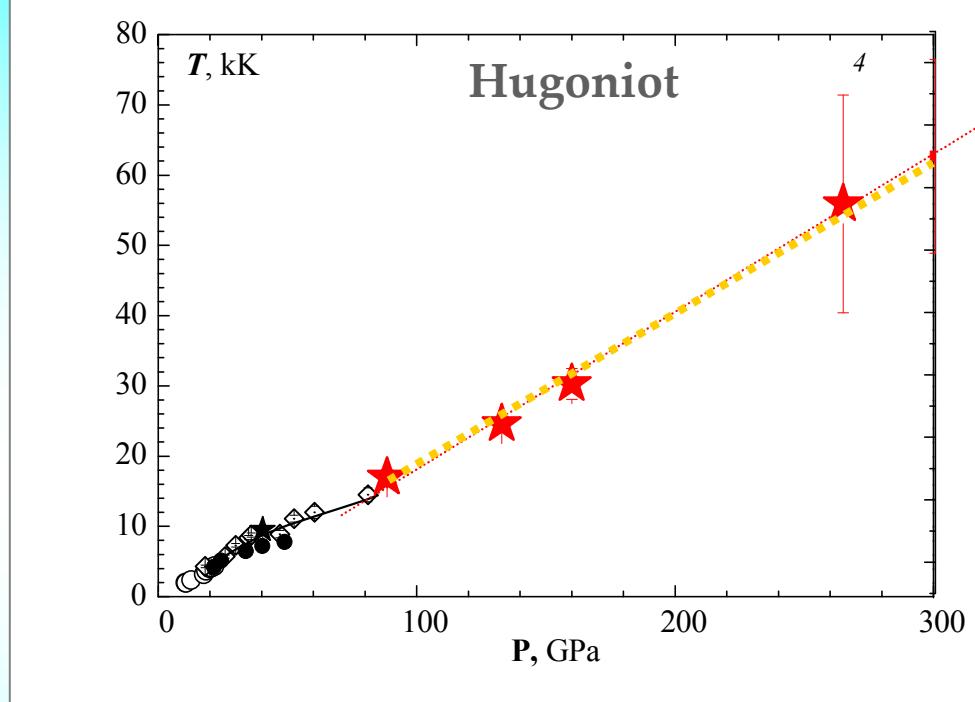
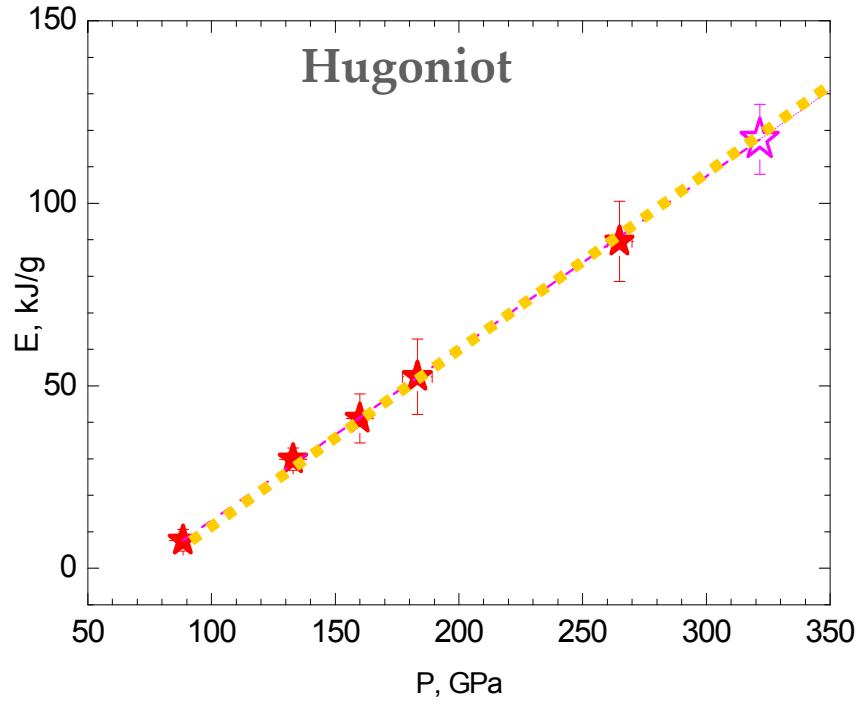
.....

High-temperature part of nitrogen Hugoniot

Internal energy ~ linear function on temperature at isochores

$$\rho \approx 3.3 \pm 0.1 \text{ g/cc}$$

$$(90 < P < 330 \text{ GPa})$$



Isochoric heat capacity (C_V) is almost constant on nitrogen isochores

NB !

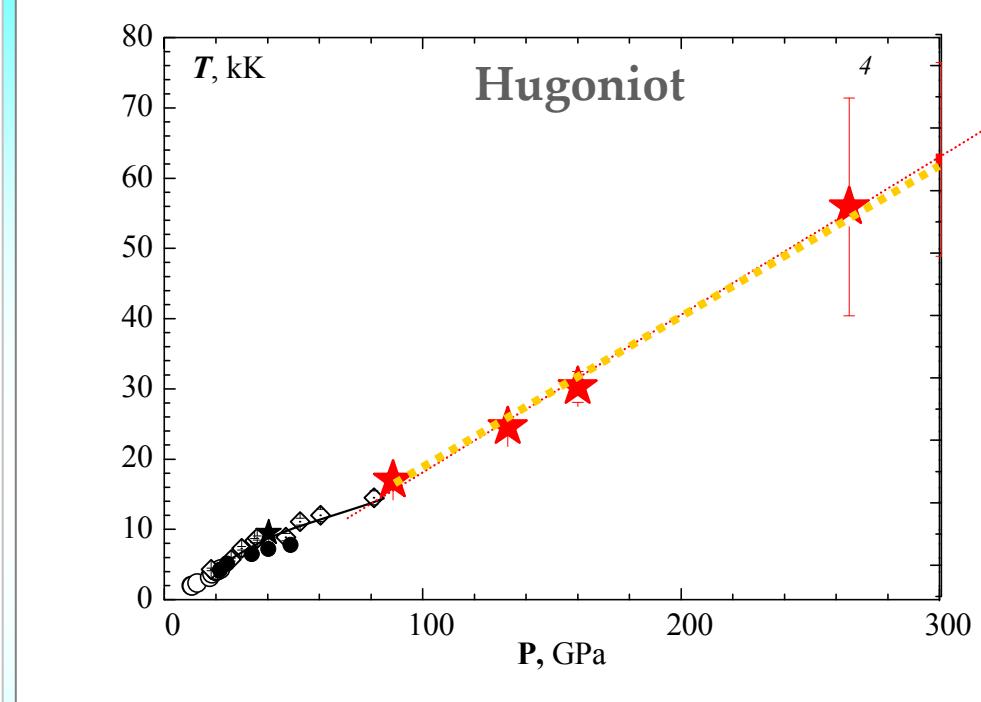
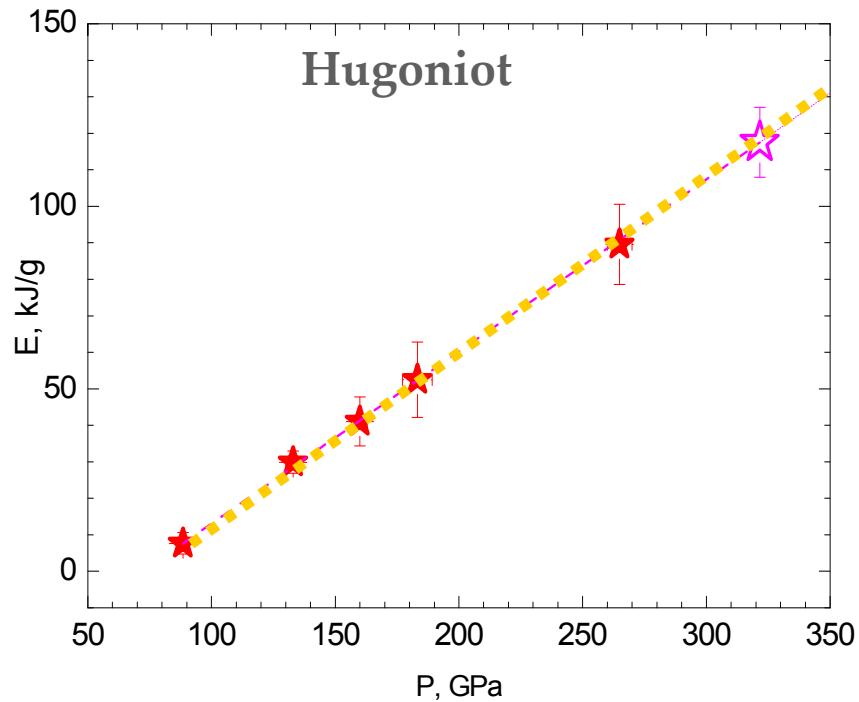
$$C_v \equiv (\partial E / \partial T)_v \approx \text{const} \approx 2.0 \text{ (J/gK)}$$

$$C_V / R_N \approx 3.5 !$$

What does it mean ?

Constancy of Grüneisen parameter and Heat capacity at isochore

$$\rho \approx 3.3 \pm 0.1 \text{ g/cc} \quad (90 < P < 330 \text{ GPa})$$



$\text{Gr} \approx \text{const} + C_V \approx \text{const} \Leftrightarrow \text{thermal pressure coefficient } \gamma_v^* \approx \text{const}$

NB !

$$(\partial p / \partial T)_v \approx 4.54 \text{ GPa/K} !$$

$$\gamma_v^* \equiv (v/k_B) (\partial p / \partial T)_v \approx \text{const} \approx$$

Summary

Thermodynamics of shock compressed nitrogen

High-temperature part nitrogen Hugoniot is close to be isochoric

$$\rho \approx 3.3 \pm 0.1 \text{ g/cc} \quad (90 < P < 330 \text{ GPa})$$

Internal energy is almost linear on pressure at isochore

$$Gr = V(\partial P / \partial E)_V \approx const \approx 0.62$$

Temperature is almost linear on pressure at isochore

$$(\partial p / \partial T)_V \approx const \approx 4.54 \text{ (GPa/K)}$$

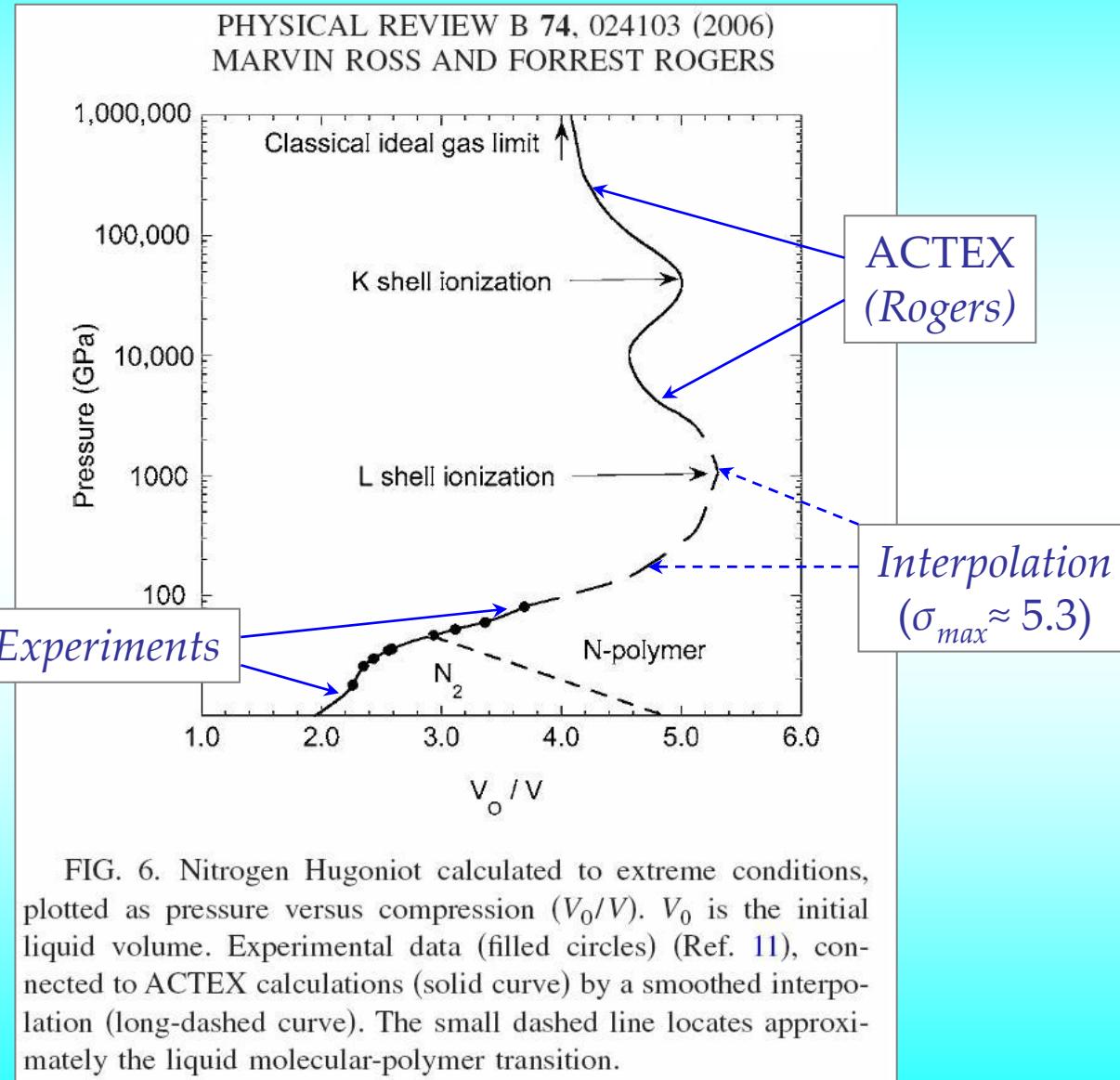
$$Z \equiv PV/RT \equiv P/n_N kT \approx const \approx 2.66 \pm 0.20$$

Internal energy is almost linear on temperature at isochore

$$C_V \equiv (\partial E / \partial T)_V \approx const \approx 2.06 \text{ (J/g·K)}$$

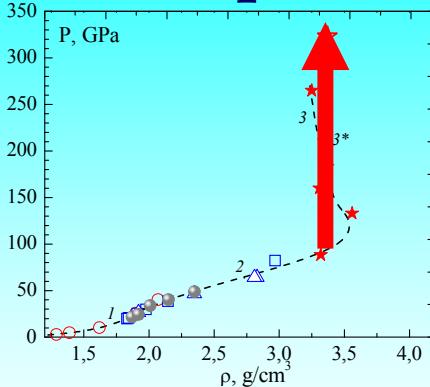
Expected behavior of nitrogen Hugoniot

M.Ross & F.Rogers



New experiments *vs* expected behavior of nitrogen Hugoniot

M.Ross & F.Rogers



???

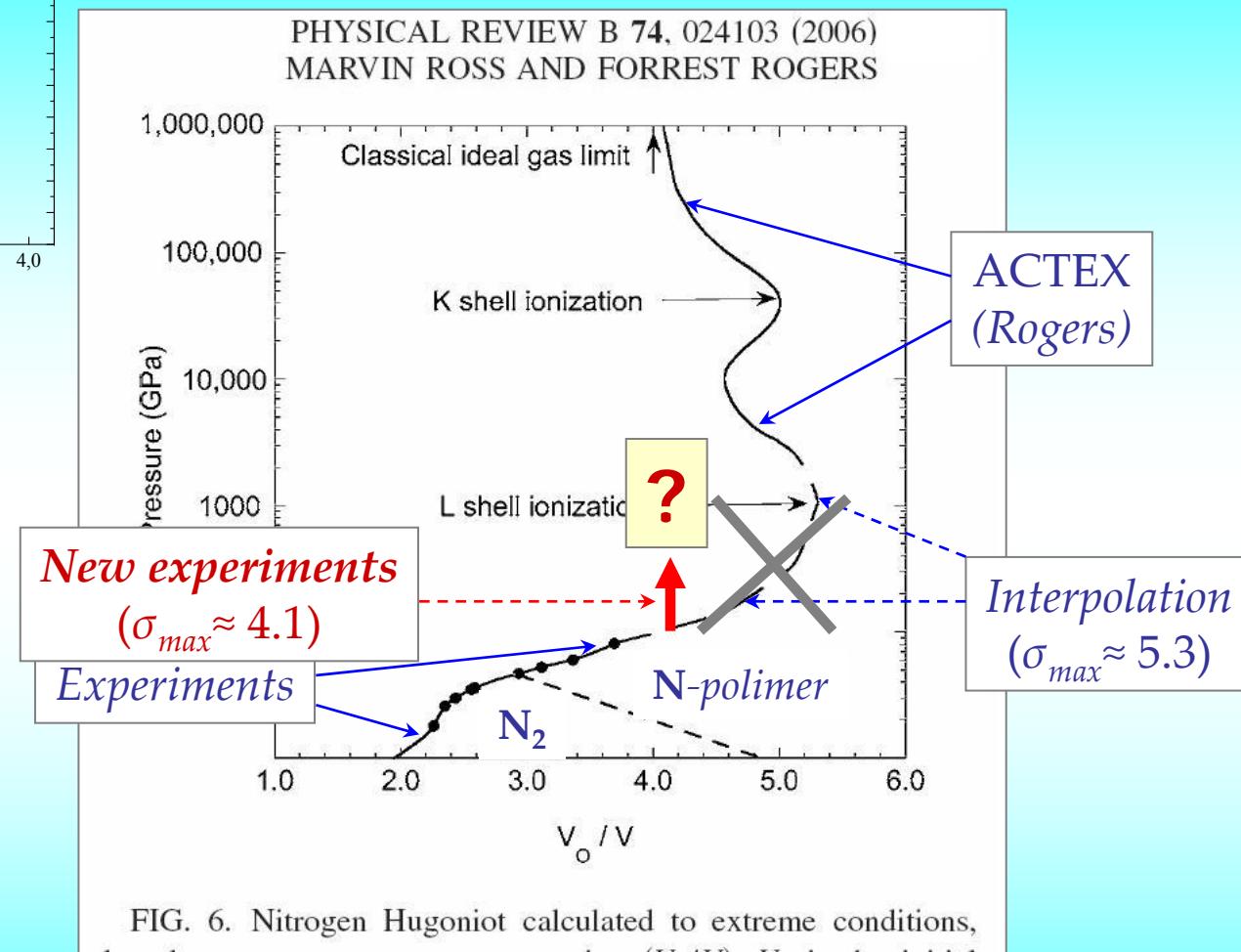


FIG. 6. Nitrogen Hugoniot calculated to extreme conditions,

NB!

New experiments *are in strong contradiction*
with expected behavior of nitrogen Hugoniot

Nitrogen Hugoniot problem in plasma state – what do we need?

New experiments ?

- Shock waves
- Isentropic Compression
- Heavy Ion Beams
- Laser Heating
- etc. etc.

Theory ?

Ab initio

RPIMC, DPIMC
DFT/MD WP/MD
Chemical models

SAHA-S

SAHA-N

(Gryaznov *et al.*)

WZ-cell models

TF, TFC, MHFS...

.....

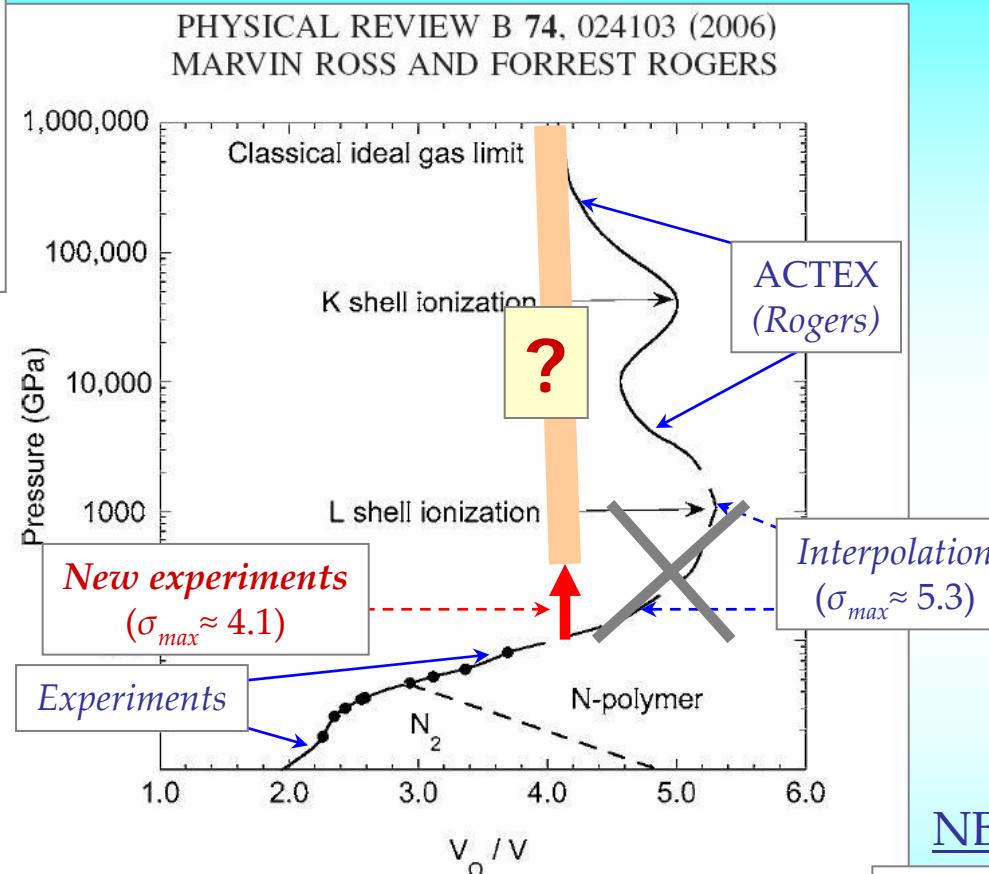
Wide-range EOS-s

IPCP RAS

JIHT RAS

CCM (Sarov)

.....



NB !

FIG. 6. Nitrogen Hugoniot calculated to extreme conditions plotted as pressure versus compression (V_0/V). V_0 is the initial liquid volume. Experimental data (filled circles) (Ref. 11), connected to ACTEX calculations (solid curve) by a smoothed interpolation (long-dashed curve). The small dashed line locates approximately the liquid molecular-polymer transition.

In last experiments
on deuterium
compression:

$$P_{max} \approx 50 \text{ Mbar}$$

(see: Mochalov *et al.* NPP-2011)
(Mochalov *et al.* JETP Lett. 2010)

Nitrogen Hugoniot

(comes through polymeric phase ?)

New experiments ?

- Shock waves
- Heavy Ion Beams
- Laser Heating
- etc. etc.

THEORY ?

Ab initio

RPIMC, DPIMC
DFT/MD, WP/MD..

Chem. models

SAHA-S

SAHA-N

.....

WZ-cell models

TFC, MHFS, ...

.....

Wide-range EOS

CCM (Sarov)

.....

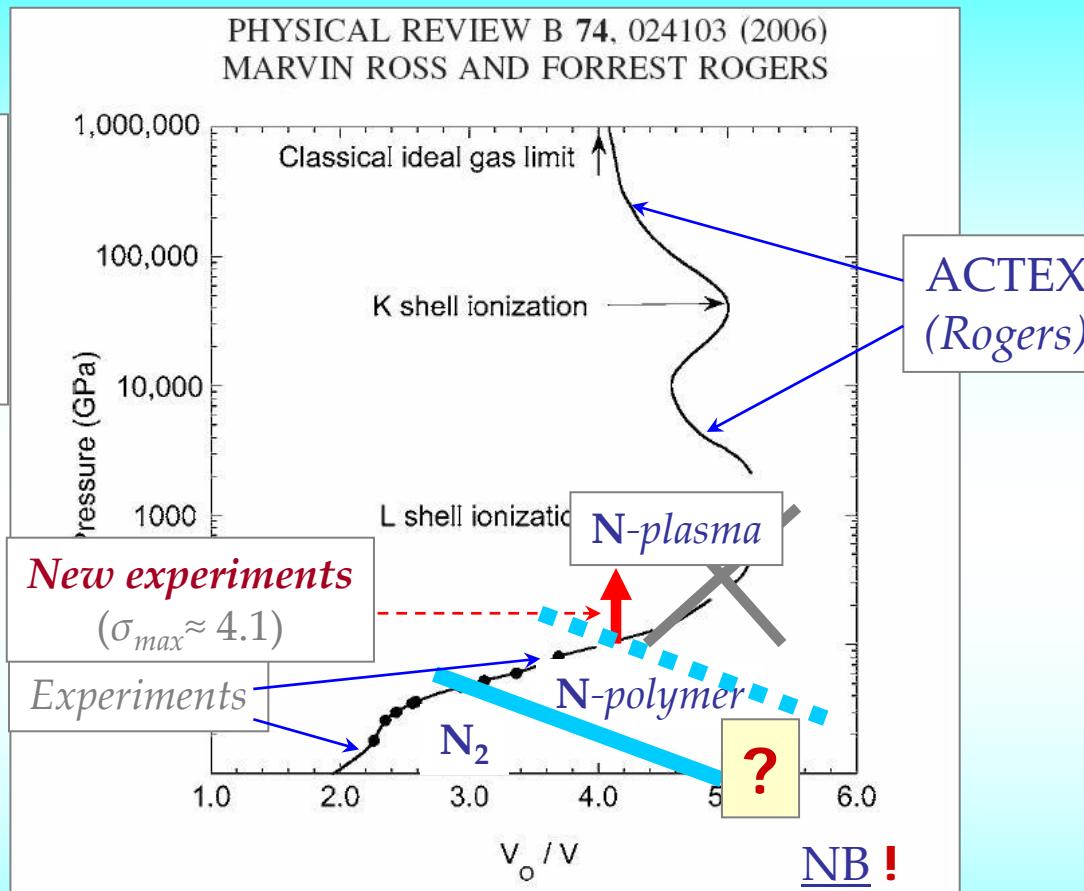


FIG. 6. Nitrogen Hugoniot calculated plotted as pressure versus compression liquid volume. Experimental data (filled circles connected to ACTEX calculations (solid curve) and long-dashed curve). The small dashed curve (short-dashed) marks the liquid molecular-polymer transition.

Polymeric state
is expected between
molecular and plasma
regions
(see below)

Ab initio calculations - DFT/MD

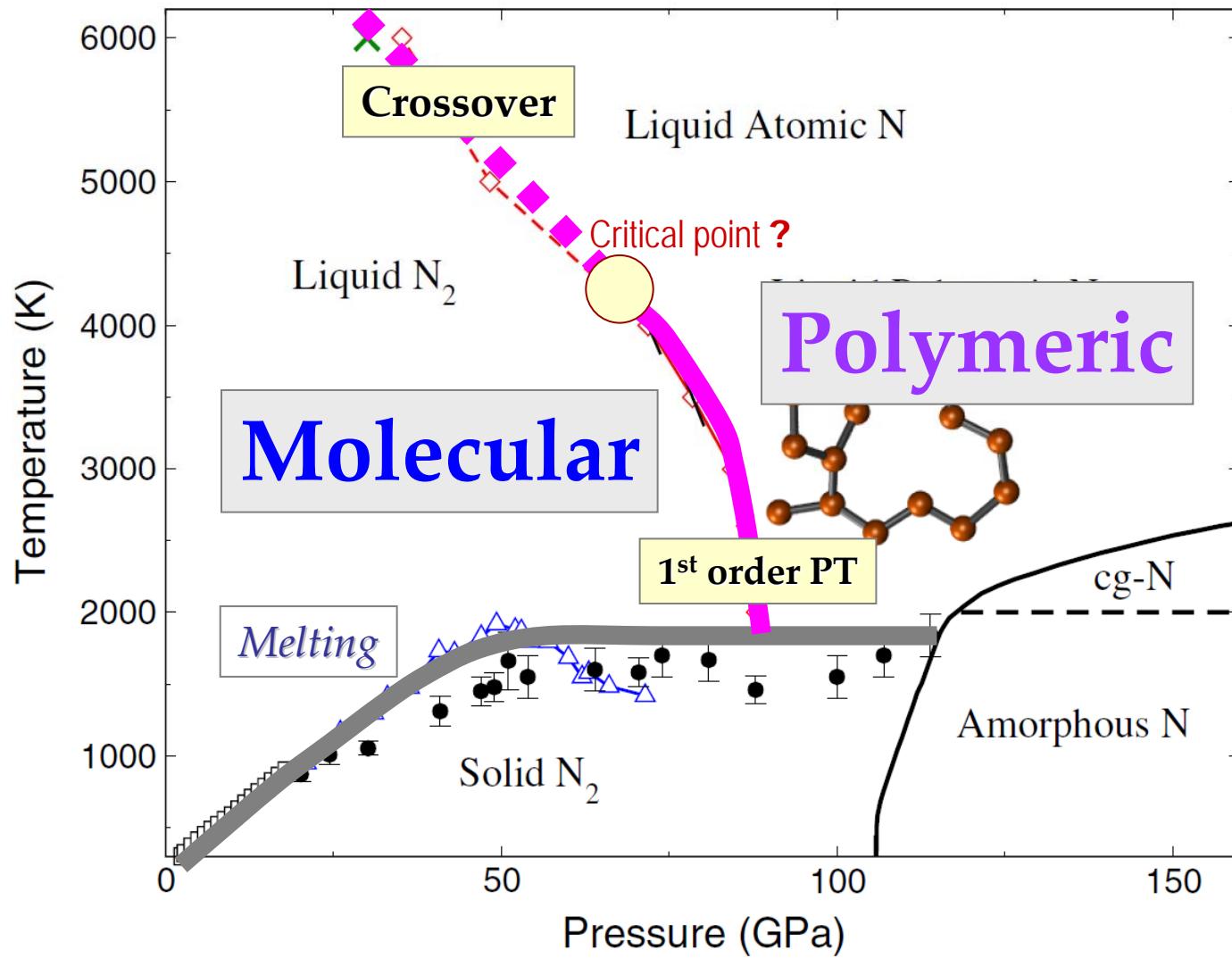


figure from
B. Boates, S. Bonev, *Phys. Rev. Lett.*, **102** (2009)

Ab initio calculations - DFT/MD

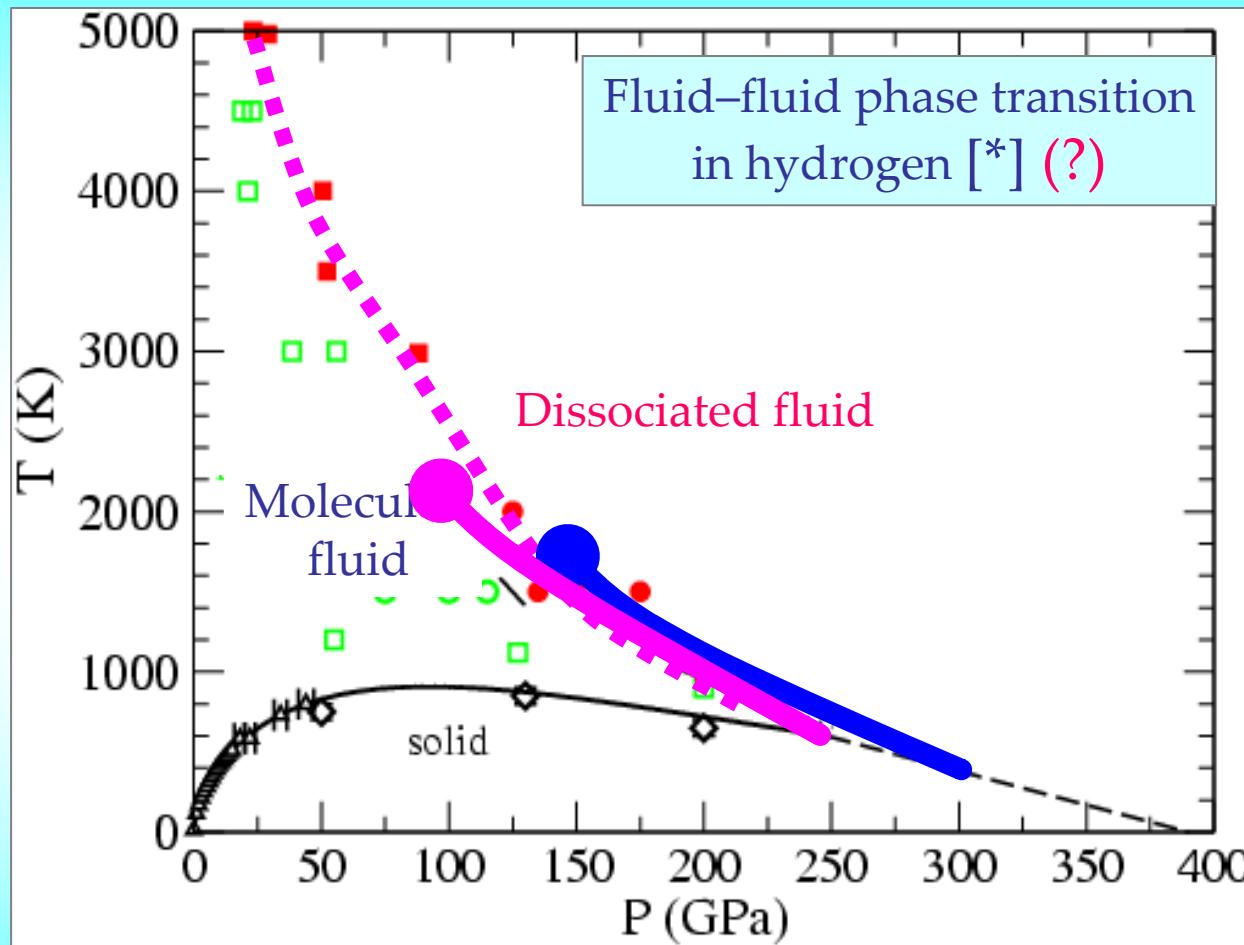
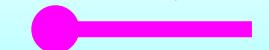


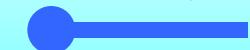
Figure from:
Giulia Galli,
SCCS-2005,
Moscow

= = @ = =
with additions:
Morales et al. 2010
and
Lorenzen et al. 2010

Morales et al, 2010

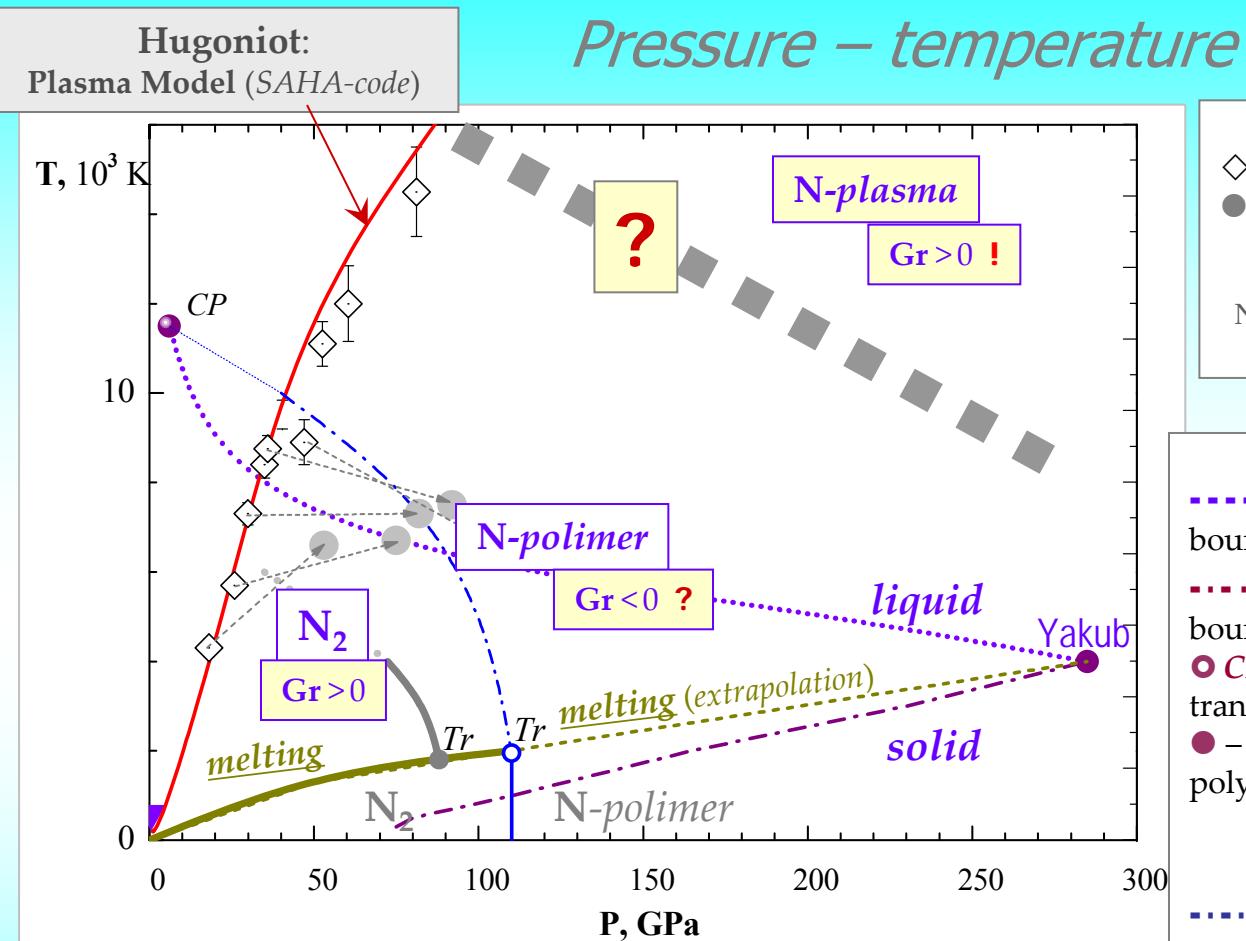


Lorenzen et al, 2010



DFT/MD: Scandolo S. *PNAS* **100**, (2003) // Bonev S., Militzer B., Galli G. *PRB* **69** (2004)
WPMD : Jakob B. *et al. PRE* (2007) // DFT/MD: Morales M. *et al. PNAS* **107**, (2010)/
DFT/MD: Lorenzen W. *et al. PRB* (2010)

Nitrogen phase diagram in the region of polymerization



$\mathbf{Gr} > 0 // \mathbf{Gr} < 0$ – domains of positive and negative sign
of Grüneisen coefficient $\{ \mathbf{Gr} \equiv V(\partial P / \partial E)_V \}$

..... hypothetical boundary between polymeric and non-ideal
plasma states (1st-order phase transition or continuous ?)

Experiment

- ◇ – single shock compression,
- – reflected shock compression

Radousky, Nellis & Ross et al,
Phys. Rev. Lett. **57** (1986)

Nellis, Radousky & Hamilton et al,
J. Chem. Phys. **94** (1991)

Theoretical models

- estimation of polymer-molecule boundary in liquid nitrogen (E.Yakub – 1993)
- estimation of polymer-molecule boundary in solid nitrogen (L.Yakub – 1993)
- CP – critical point of 1st order phase transition polymer-molecule (E.Yakub – 1993)
- – triple point of 1st order phase transition polymer-molecule (E. & L. Yakub – 1993)

E.S. Yakub, / *Low Temp. Phys.* **20** 579 (1994)/
L.N. Yakub, / *Low Temp. Phys.* **19**, 531 (1993)/

- estimation of polymer-molecule boundary in liquid nitrogen (Ross & Rogers – 2006)

M.Ross & F.Rogers, *Phys. Rev. B* **74** (2006)

Ab initio calculations

- DFT/MD – calculation of polymer-molecule boundary in liquid nitrogen (S.Bonev et al. – 2009)
- the same as “smoothed” phase transition

B. Boates & S. Bonev, *Phys. Rev. Lett.* **102** (2009)

Phase Diagram of Dense Nitrogen

(summary)

New experiments ?

- Shock waves
- Iso-S Compression
- Heavy Ion Beams
- Laser Heating
- etc. etc.

THEORY ?

Ab initio

RPIMC, DPIMC
DFT/MD

Chem. models
SAHA-N

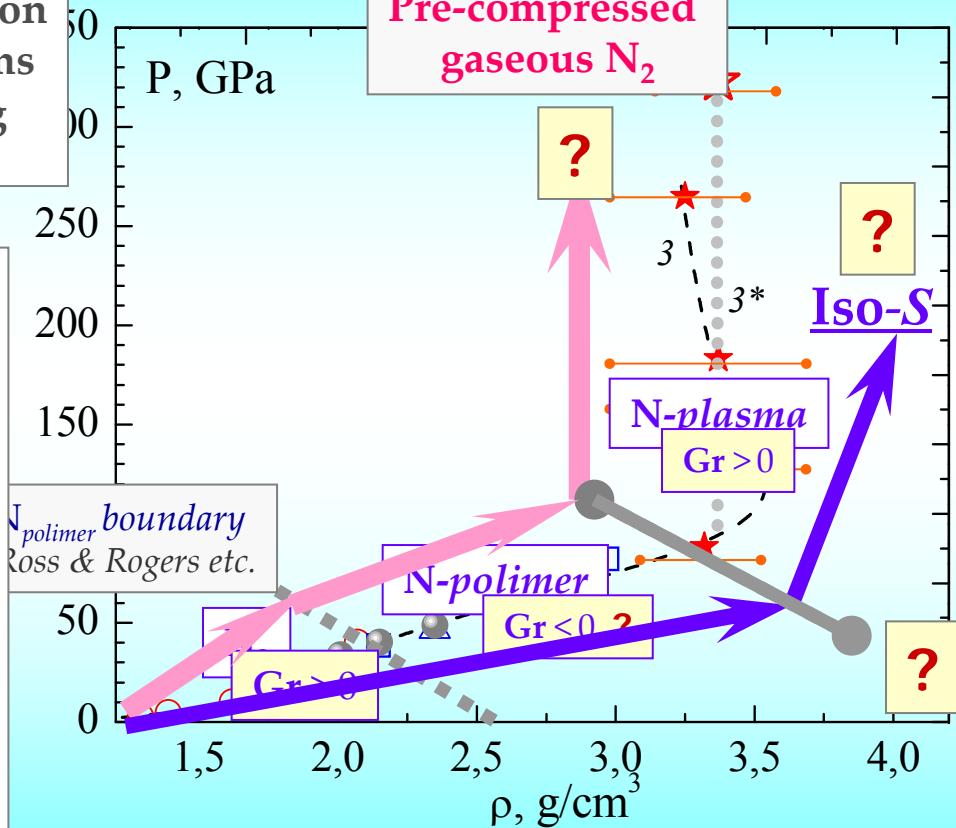
.....

WZ-cell models
TFC, MHFS, ...

.....

Wide-range EOS
EOS (IPCP RAS)
EOS (JIHT RAS)
CCM (Sarov)

.....



Hypothetical Pressure Ionization from N-polymer to N-plasma

1st-order phase transition ?
Critical point(s) ?

Domain: Gr < 0 ?
Topology of the boundary: Gr = 0 ?



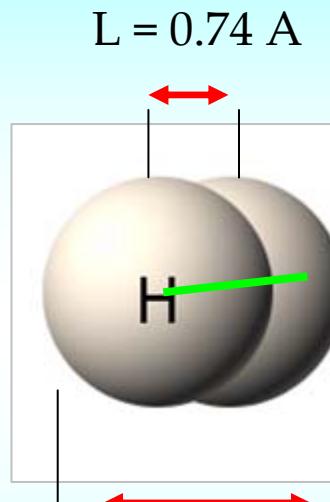
Comments

microphysics

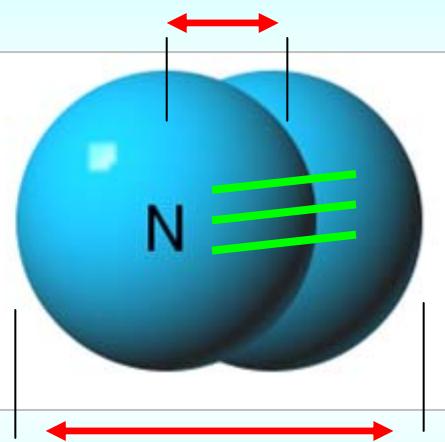
Simple molecules: $\text{H}_2 \leftrightarrow \text{N}_2$

Triple covalent bond

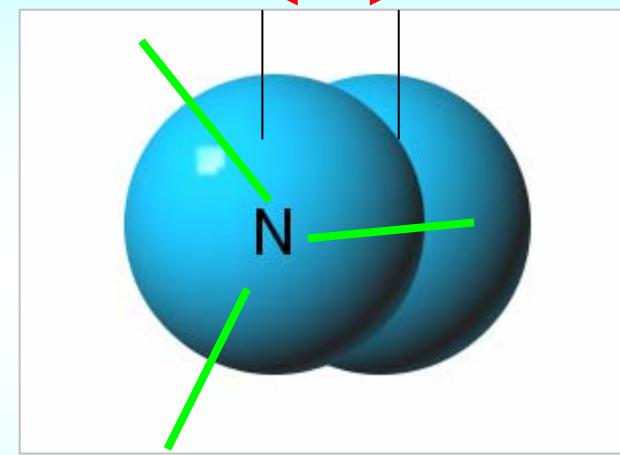
$$D_e = 9.8 \text{ eV} \quad (4.9 \text{ eV/atom})$$



$$L = 1.1 \text{ \AA}$$



$$L = 1.4 \text{ \AA}$$



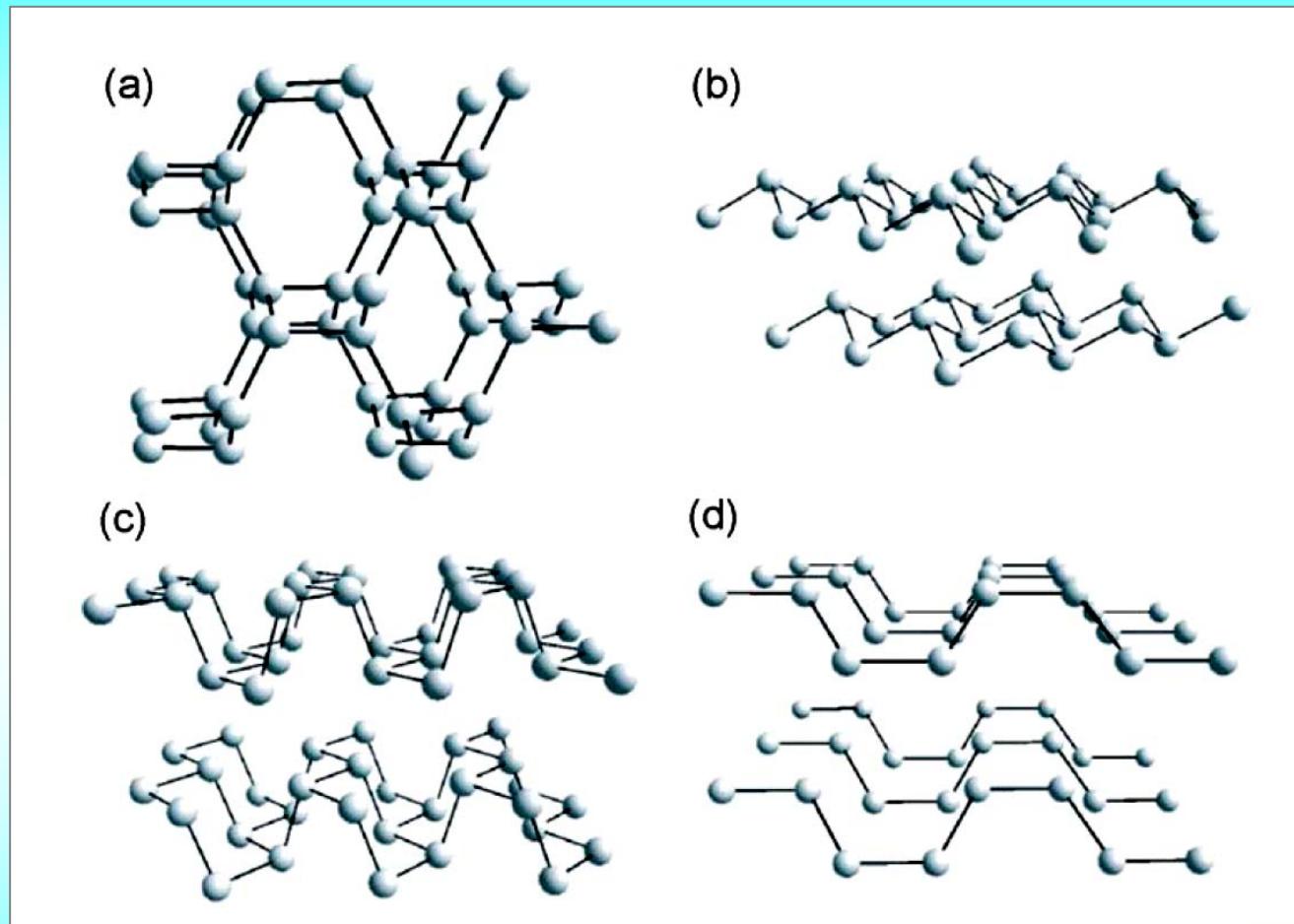
Single covalent bond

$$D_e = 3.0 \text{ eV} \quad (4.5 \text{ eV/atom})$$

Figures after Eugene Yakub: "*Non-simple problems for simple molecules*"

FAIR-Russia School «*Physics of high energy density in matter*» December 2009, Moscow

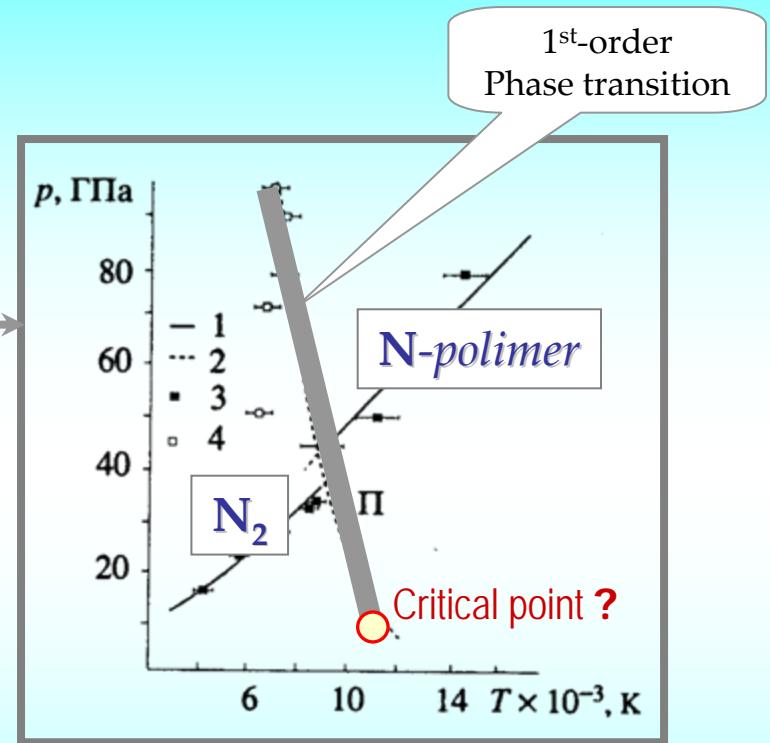
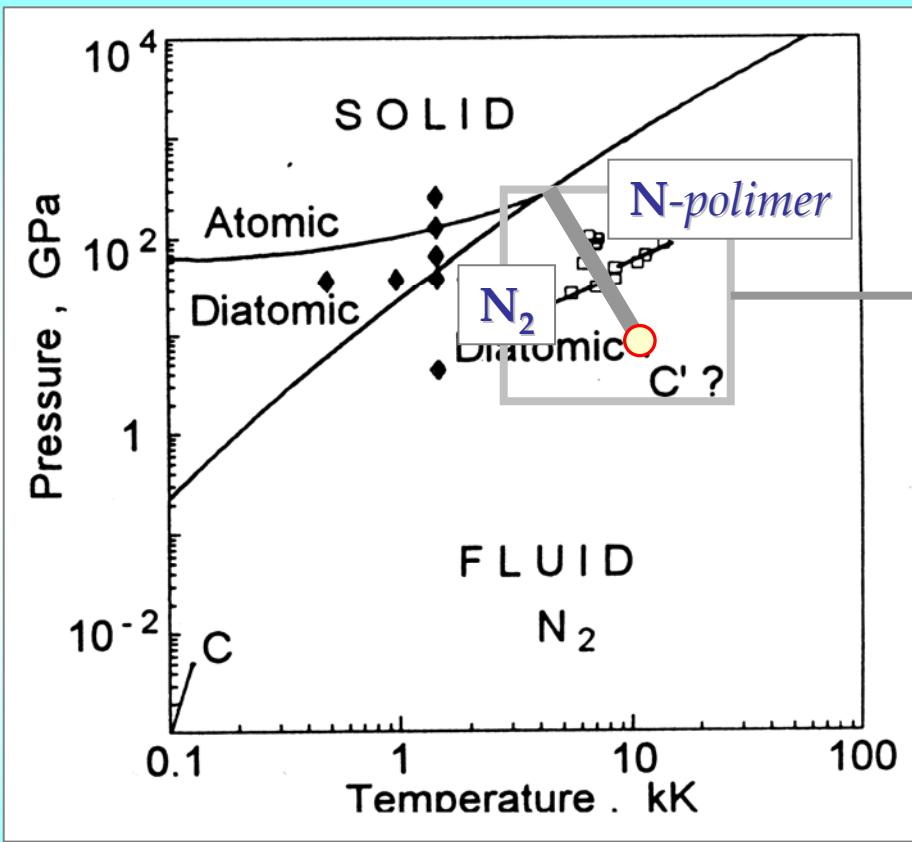
Polymeric nitrogen - structure



F.Zahariev,A.Hu,J.Hooper,F.Zhang,&T.Woo. Phys.Rev. B 72,214108 (2005)

Simple molecular models

First estimations of molecular-polymeric transition



Eugene *and* Lidia Yakub, *Low Temp. Phys.* 19, (1993)

- 1 - Hugoniot
- 2 - Phase equilibrium line
Experiment (Nellis ea, 2001)
- 3 - Single shock
- 4 - Double shock

Plasma model for nitrogen thermodynamics

(code SAHA-N)

Shock compression of nitrogen

(chemical picture)

SAHA-code

Gryaznov, Iosilevskiy (1970-2010)

Equilibrium composition:

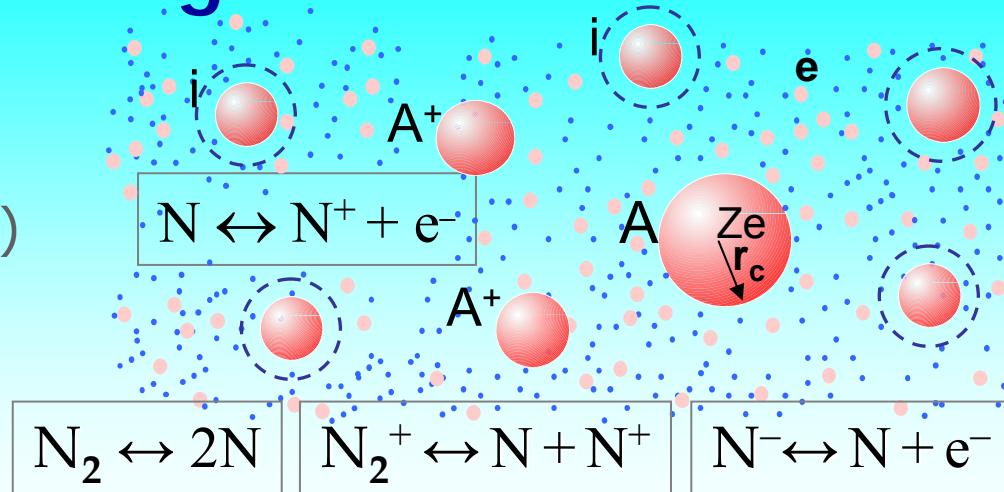
$\{\text{N}_2, \text{N}, \text{N}^+, \text{e}, \text{N}_2^+, \text{N}^-\}$

Coulomb interaction:

Modified pseudopotential model
for partially ionized plasma
(I. Iosilevskiy 1980-2010)

Short range repulsion:

“Soft Spheres” approximation (D. Young)
modified for mixture of soft spheres with
different radii – shift of dissociation level.



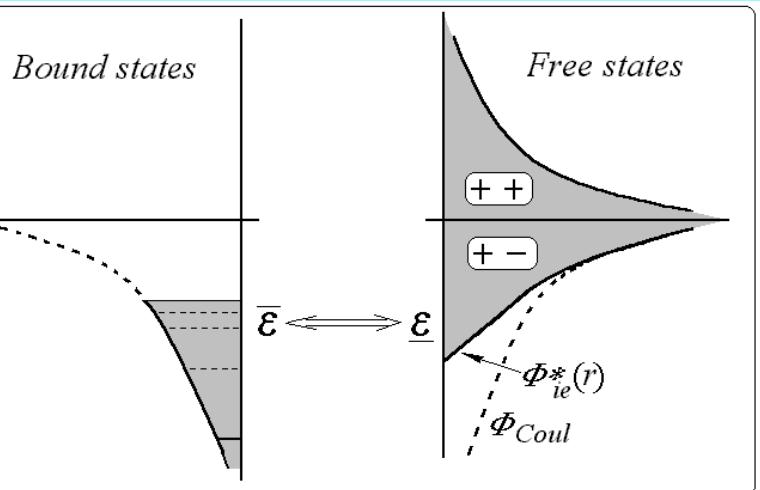
Basic points:

- All assumption are provided at micro-level.
- Input: Choice of Φ_{ie} , Φ_{ii} and Φ_{ee} pseudopotentials.
- Input: Approximations for binary correlation functions
- Strong correlation of the pseudopotentials for “free” charges and upper energy level for partition functions for bound states.
- Priority for general equalities (*normalizing conditions*) valid independently on degree of non-ideality.
- Non-ideality corrections through the correlation functions and general equalities, valid for Coulomb interaction.

Key parameter – ratio of intrinsic volumes: *molecule / atom / ion*: $\mathbf{R(N) / R(N_2) = 0.63}$
- in accordance with recommendations of “Atom-atomic approximation” of E. Yakub, LT, 1993

Modified Pseudopotential Approximation (*):

Bound and free states of electron-ionic pair



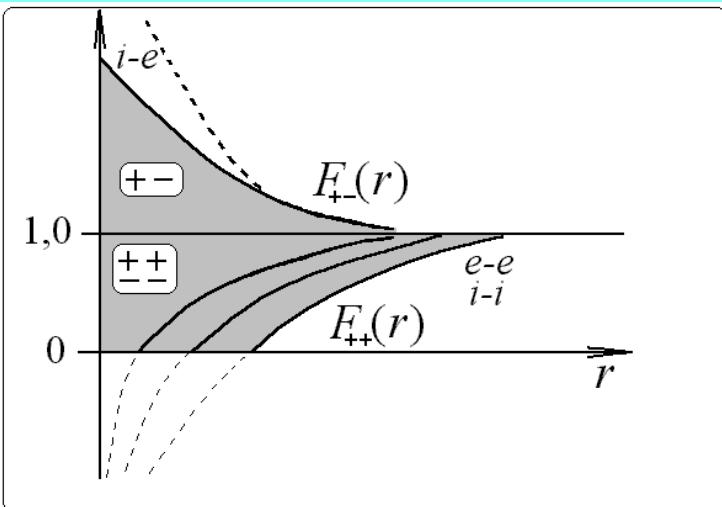
Effective electro-ionic potential
(in Glauberman's form (1955))

$$\Phi_{ie}^*(r) = -\frac{Z_i e^2}{r} (1 - e^{-r/\sigma}) \geq -\varepsilon \quad \Leftrightarrow$$

Thermodynamic condition for the depth of i - e pseudopotential
and amplitude of electron-ionic correlation function

$$-\Psi_0 \approx -\ln F_{ie} \approx \beta[\Phi_{ei}^*(0) - \Delta\mu_e - \Delta\mu_i]$$

Binary i - i , i - e , e - e correlation functions



The form of i - i , i - e , e - e binary correlation functions
- «ring» (Debye) approximation with potential $\Phi_{ie}^*(r)$

$$F_{ab}(r) = 1 \pm A \frac{e^{-pr} - e^{-qr}}{r} \equiv 1 \pm \Psi_0 e^{-vr} \frac{\sinh\{\omega r\}}{\omega r}$$

"Zero and second moment" conditions (Stillinger & Lovett)

$$n \int [F_+(r) - F_-(r)] d\mathbf{r} = 1$$

$$n \int [F_+(r) - F_-(r)] \left(\frac{r^2}{r_D^2} \right) d\mathbf{r} = 3$$

Positive sign of all correlation functions

$$\Leftrightarrow F_{ab}(r) > 0$$

Thermodynamic contributions in modified pseudopotential model for Coulomb corrections

$$\Phi_{ie}^*(r) = -\frac{Z_i e^2}{r}(1 - e^{-r/\sigma}) \equiv -\left(\frac{Z_i e^2}{\sigma}\right)\frac{(1 - e^{-r/\sigma})}{r/\sigma} \geq -\Phi_{ie}^*(0) \sim -\epsilon$$

Pseudopotentials

$$F_{ab}(r) = 1 \pm A \frac{e^{-pr} - e^{qr}}{r} \equiv 1 \pm \Psi_0 e^{-vr} \frac{\sin \omega r}{\omega r}$$

Correlation Functions

Homogeneity of Coulomb potential \Leftrightarrow

$$U = U_{Kin} + U_{Pot} \quad 3PV = 2U_{Kin} + U_{Pot}$$

Total Energy correction

$$\Delta U = -Vn^2 \int (F_+ \Phi_{ei}^* - F_- \Phi_{ii}^*) d\mathbf{r}$$

Potential Energy Correction - ΔU_{Pot}

$$\Delta U_{pot} = -Vn \int \Phi_{coul} (F_+ - F_-) d\mathbf{r}$$

Pressure Correction - ΔP

$$3\Delta PV = (2\Delta U - \Delta U_{pot}) = (2\Delta U_{kin} + \Delta U_{pot})$$

Approximate relation between Coulomb corrections for chemical potential and energy ($\Delta\mu \Leftrightarrow \Delta U/N$)

$$\Delta\mu_i = \Delta\mu_e \approx (N_i + N_e)^{-1} \Delta U$$

NB !

Positive shift in average kinetic energy due to non-ideality of free charges subsystem

$$\Delta U_{kin} = 3\Delta PV - \Delta U$$

Well-known relation between pressure and energy corrections for free charges subsystem

$$\Delta U = 3\Delta PV$$

which is valid at weak non-ideality ($\Gamma \ll 1$), is no longer valid in strong non-ideality conditions ($\Gamma \sim 1$)

It is equivalent to **additional effective electron-ion repulsion**

(in comparison with ordinary one-parametric Coulomb corrections, depending on non-ideality parameter only)

$$\Delta F/NkT = f(\Gamma)$$

Experimental data \Leftrightarrow Theoretical models

(*comparison*)

Summary

Thermodynamics of shock compressed nitrogen *(primary thermodynamic results of experiment)*

High-temperature part of Hugoniot is close to isochore

$$\rho \approx 3.3 \pm 0.1 \text{ g/cc} \quad (90 < P < 330 \text{ GPa})$$

Internal energy is almost linear on pressure at isochore

$$Gr = V(\partial P / \partial E)_V \approx const \approx 0.62$$

Temperature is almost linear on pressure at isochore

$$(\partial p / \partial T)_V \approx const \approx 4.54 \text{ (GPa/K)}$$

$$Z \equiv PV/RT \equiv P/n_N kT \approx const \approx 2.66 \pm 0.20$$

Internal energy is almost linear on temperature at isochore

$$C_V \equiv (\partial E / \partial T)_V \approx const \approx 2.06 \text{ (J/g·K)}$$

Check of theoretical models

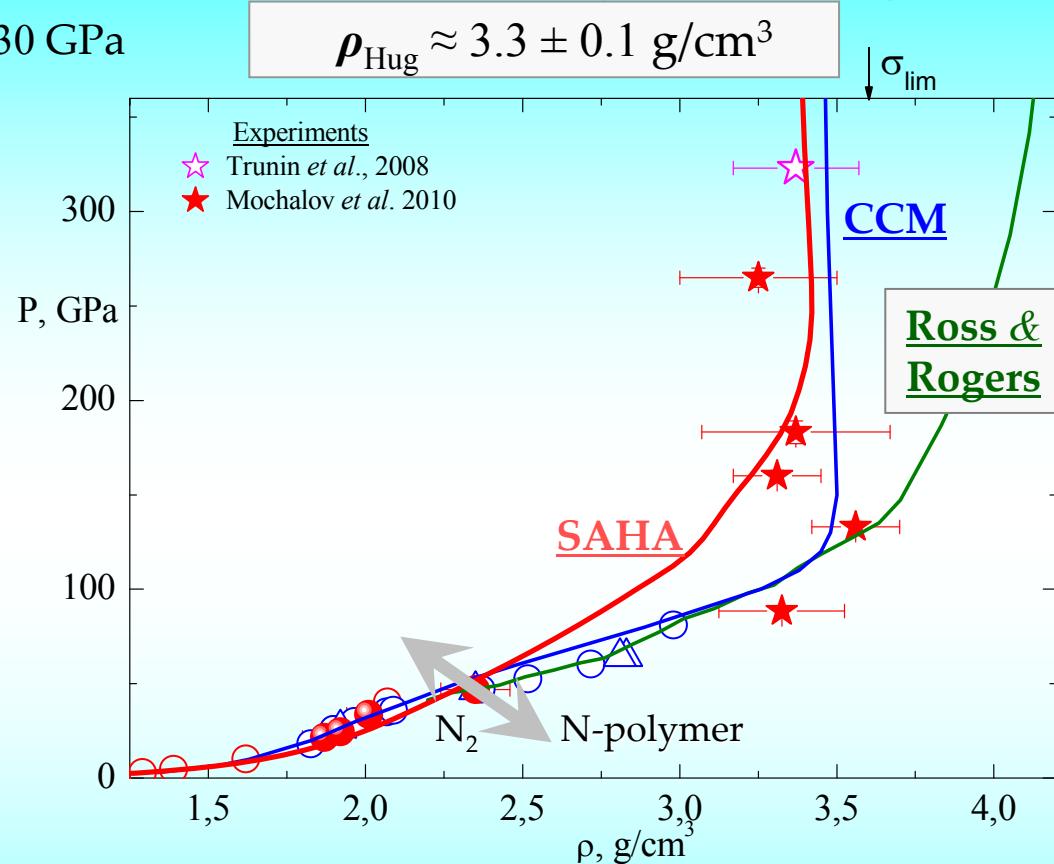
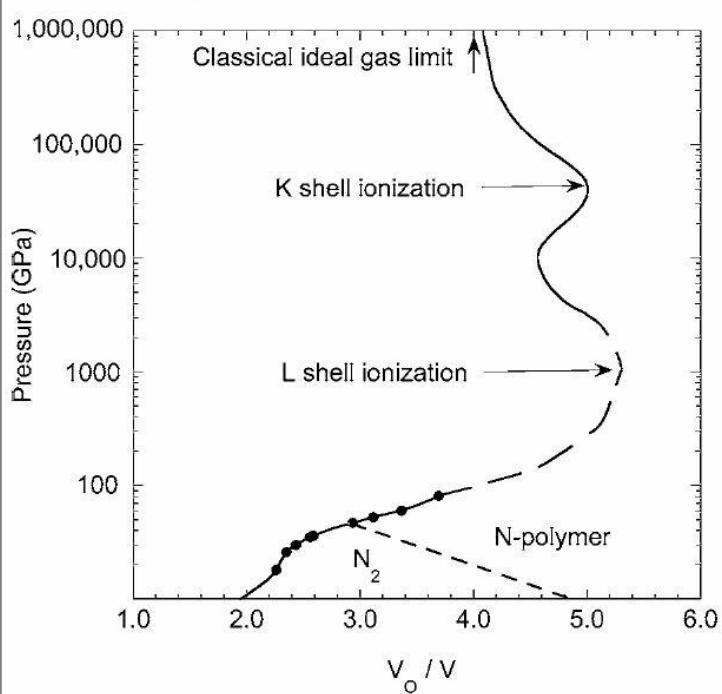
Quasi-isochoric behavior of nitrogen Hugoniot

$$90 < P_{\text{Hug}} < 330 \text{ GPa}$$

$$\rho_{\text{Hug}} \approx 3.3 \pm 0.1 \text{ g/cm}^3$$

 $\downarrow \sigma_{\text{lim}}$

PHYSICAL REVIEW B 74, 024103 (2006)
MARVIN ROSS AND FORREST ROGERS



SAHA-code (*V. Gryaznov & I.Iosilevskiy*)

CCM – Compressible Covolume Model
(*A. Medvedev & V.Kopyshev*)

★ - Mochalov M., Zhernokletov M. et al.,
JETP, 137 (2010)

★ - Trunin R., Boriskov G., et al.
JETP Letters, 88 (2008)

Summary

Thermodynamics of shock compressed nitrogen (primary thermodynamic results of experiment)

High-temperature part of Hugoniot is close to isochore

$$\rho \approx 3.3 \pm 0.1 \text{ g/cc} \quad (90 < P < 330 \text{ GPa})$$

Internal energy is almost linear function of pressure at isochore

$$\text{Gr} \equiv V(\partial P / \partial E)_V \approx \text{const} \approx 0.62$$

Temperature is almost linear on pressure at isochore

$$(\partial p / \partial T)_V \approx \text{const} \approx 4.54 \text{ (GPa/K)}$$

$$Z \equiv PV/RT \equiv P/n_{\text{N}}kT \approx \text{const} \approx 2.66 \pm 0.20$$

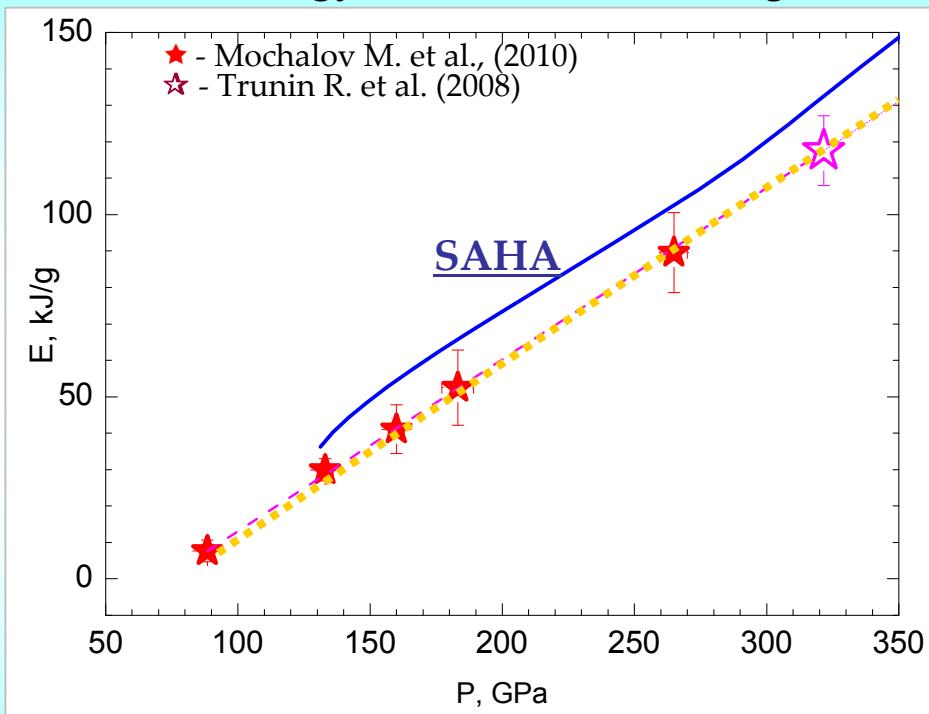
Internal energy is almost linear on temperature at isochore

$$C_V \equiv (\partial E / \partial T)_V \approx \text{const} \approx 2.06 \text{ (J/g·K)}$$

Quasi-linear behavior of $E(p)$ at $\rho=const$

Hugoniot: $\sim \rho \approx const$ (3.3 ± 0.1 g/cc)

Internal Energy \Leftrightarrow Pressure (Hugoniot)



$\rho \approx 3.3 \pm 0.1$ g/cc
 $90 < P < 330$ GPa

Blue curve – internal energy calculated via plasma model
 (code SAHA / Gryaznov, Iosilevskiy)

NB ! (*) Reference point – energy of ideal atomic gas N at $T = 0$ K

→ $Gr = V(\partial P / \partial E)_V \approx const \approx 0.62$

Summary

Thermodynamics of shock compressed nitrogen (primary thermodynamic results of experiment)

High-temperature part nitrogen Hugoniot is close to be isochoric

$$\rho \approx 3.3 \pm 0.1 \text{ g/cc} \quad (90 < P < 330 \text{ GPa})$$

Internal energy is almost linear on pressure at isochore

$$Gr = V(\partial P / \partial E)_V \approx const \approx 0.62$$

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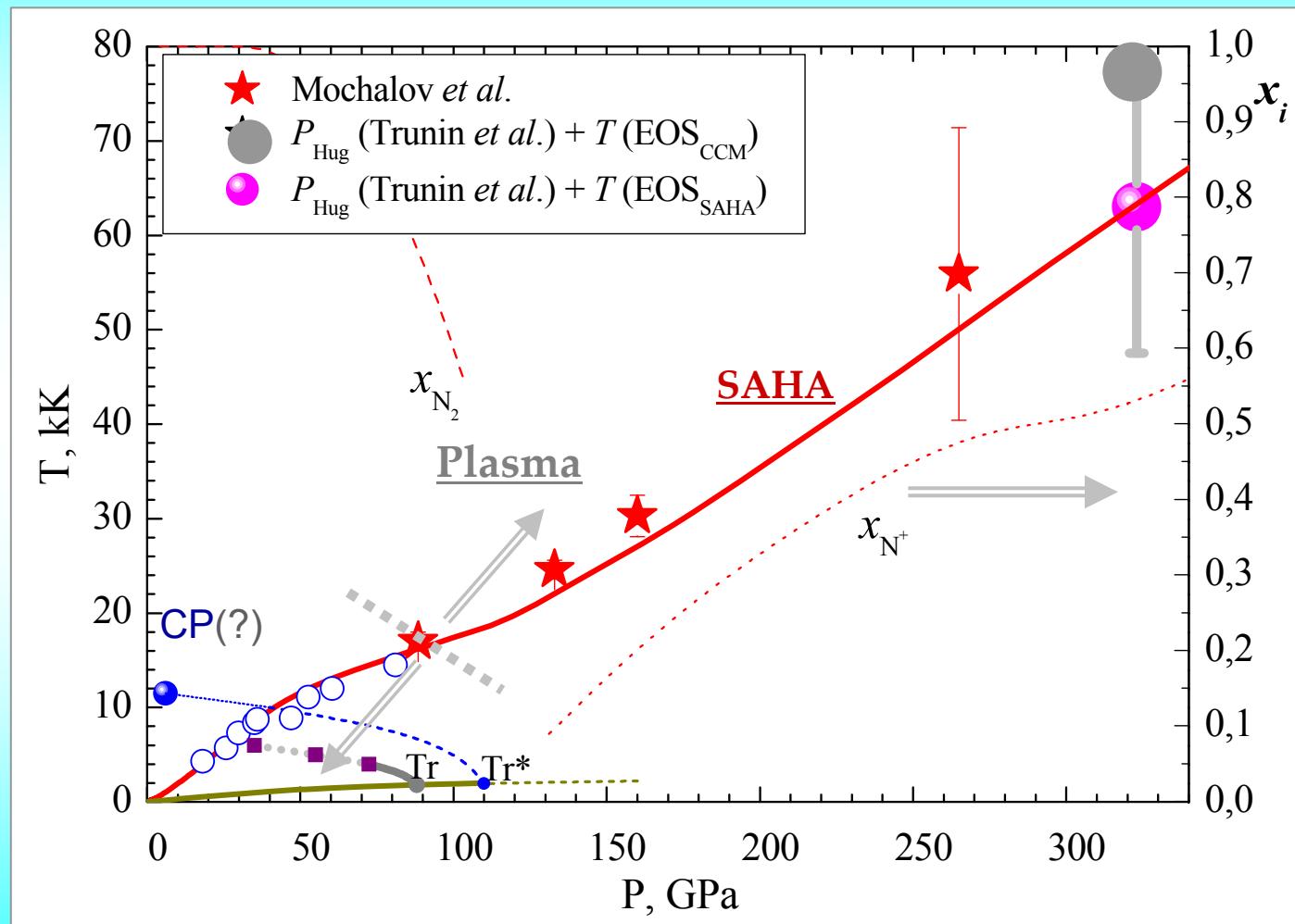
$$Z \equiv PV/RT \equiv P/n_N kT \approx const \approx 2.66 \pm 0.20$$

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$$C_V \equiv (\partial E / \partial T)_V \approx const \approx 2.06 \text{ (J/g·K)}$$

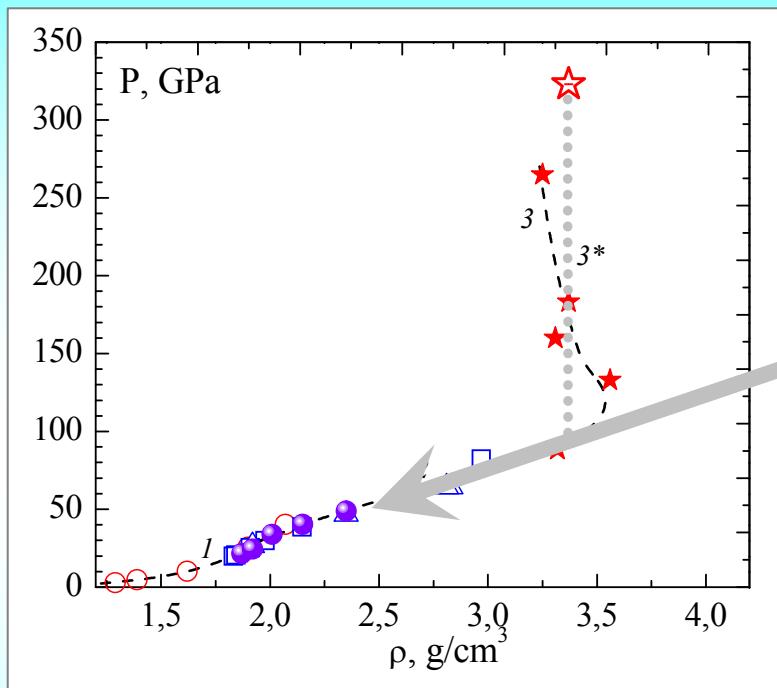
Temperature of shock compressed nitrogen

General P-T diagram

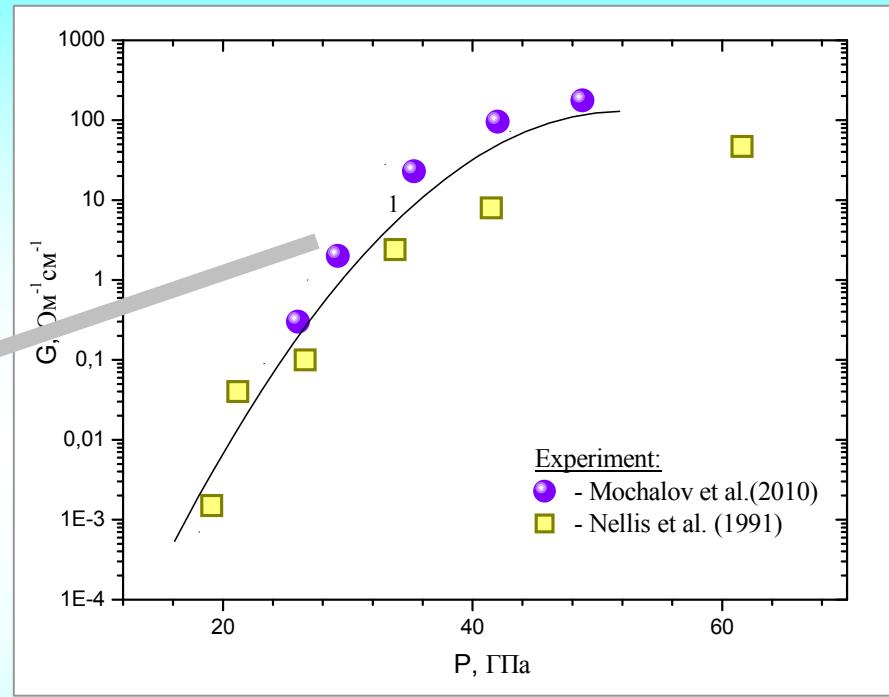


Quasi-linear behavior of $T(p)$ at $Q=const$

Electroconductivity of shock compressed nitrogen

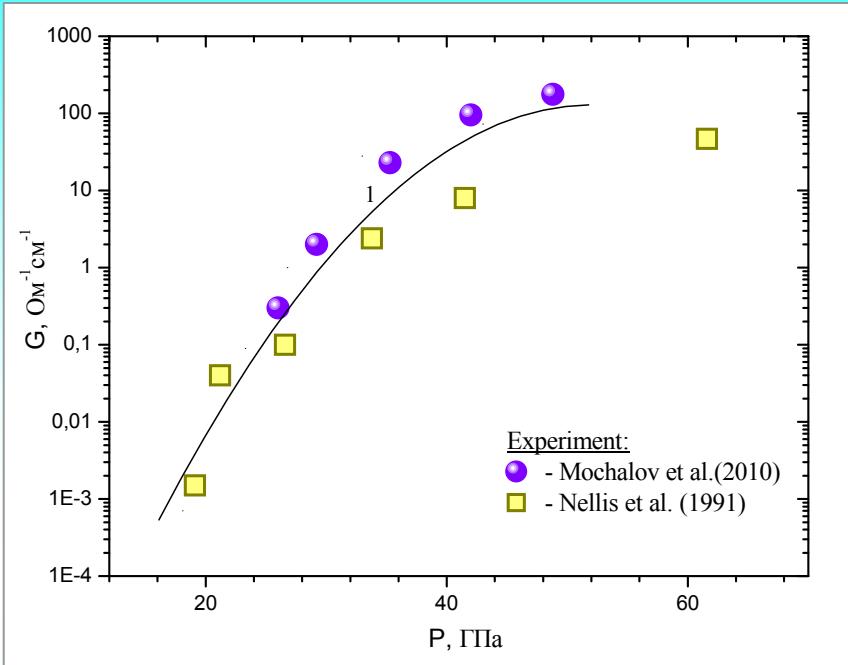


● ★ - Experiment /Mochalov M. et al. /
JETF 137 (2010)

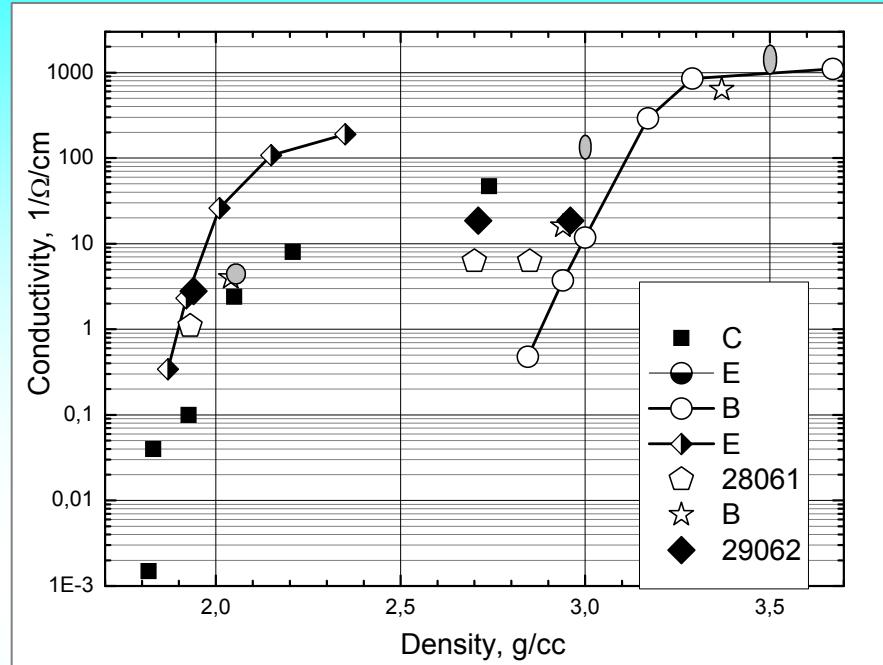


Nellis W., Radovsky H. et al. *J. Chem. Phys.* **94** (1991)

Electroconductivity of shock compressed nitrogen



● ★ - Experiment /Mochalov M. et al. /
JETF 137 (2010)



Ternovoi V. et al. (email - 30.05.2010)

Quasi-isentropic Plane Compression of Matter at Megabar Pressures by Using of a Layered System to Diminish First Shock Wave Intensity

Ternovoi V., Pyalling A., Filimonovet A. (05.2010)

Summary and conclusion

Explosive driven quasi-isentropic compression generators were proposed for matter investigation in the megabar pressure region. Results of the first experiments on quasi-isentropic compression of liquid nitrogen are presented. It was shown, that **pressure ionization** of nitrogen proceeds at **densities** from **3.15 to 3.4 g/cc** at a **temperature** of about **3000 K**. Diminishing of temperature growth was measured during onset of nitrogen electrical conductivity.

Perspectives

Modelling

Improvement of Plasma model (SAHA-code):

- *From EOS of soft spheres to EOS of {exp - 6} potential system*

Improvement of Polymeric models (E. & L. Yakub):

- *Calibration of both model on results of ab initio calculations (DFT/MD)*

Collaboration (Gryaznov, Iosilevskiy \leftrightarrow E. & L. Yakub)

- *Incorporation of polymeric state model into SAHA-code*

New approaches are desirable: (*new calculations and comparisons*)

- *Ab initio:* RPIMC, DPIMC // DFT/MD // WPMD // TBM . . .
- *Wigner-Zeits cell model:* TFC, MHFS, . . .
-
- *Semiempirical (wide-range) EOS-s*
-

Perspectives

(new experiments)

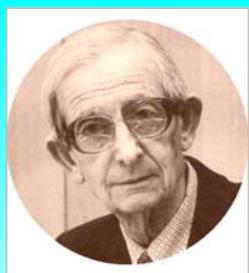
- ! - Strong **shock compression** of **liquid** nitrogen at high pressure ($P > 3$ Mbar)
- ! - Strong **shock compression** of **solid** nitrogen at high pressure ($P > 3$ Mbar)
- ! - Strong **shock compression** of **pre-compressed gaseous** nitrogen at high pressure (*mesurement of series of Hugoniots with varying initial densities (like in VNIIEF experiments with deuterium)*)
- ! - Strong **isentropic compression** of nitrogen by explosive at Mb pressure
search of density discontinuity (hypothetical phase transition ?) on nitrogen isentrope(s) (like in VNIIEF's experiments with deuterium (M.Mochalov, V.Fortov et al. PRL, 2007)

Exotics:

- ! - Heavy Ion Beam and Laser heating of cryogenic nitrogen
-

Conclusions

- New experimental data on shock compression of cryogenic liquid nitrogen in megabar pressure range “open new page” in investigation of properties for warm dense matter of “simple” molecular gases
- Simultaneous measurement of caloric and thermal equation of state (EOS) (*pressure, density, temperature and internal energy*) on the same Hugoniot give powerful tool for checking theoretical models and “calibration” of wide-range EOS-s
- New experimental data in nitrogen may be considered as the thermodynamic manifestation of non-standard form of pressure ionization (*from polymeric to plasma state*)
- This new form of pressure ionization (*from polymeric to plasma state*) seems to be general, universal and interesting phenomenon
- It is promising to continue and extend experimental investigation of pressure ionization of polymeric state
- It is promising to study pressure ionization of polymeric state in direct numerical simulations (“numerical experiment”) DFT_MD, PIMC, WP_MD...

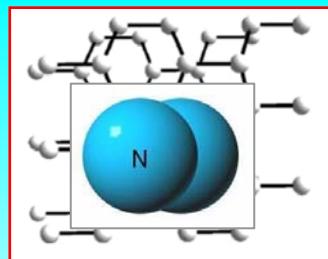


Annual Moscow Workshop

Non-ideal Plasma Physics

Russian Academy of Science

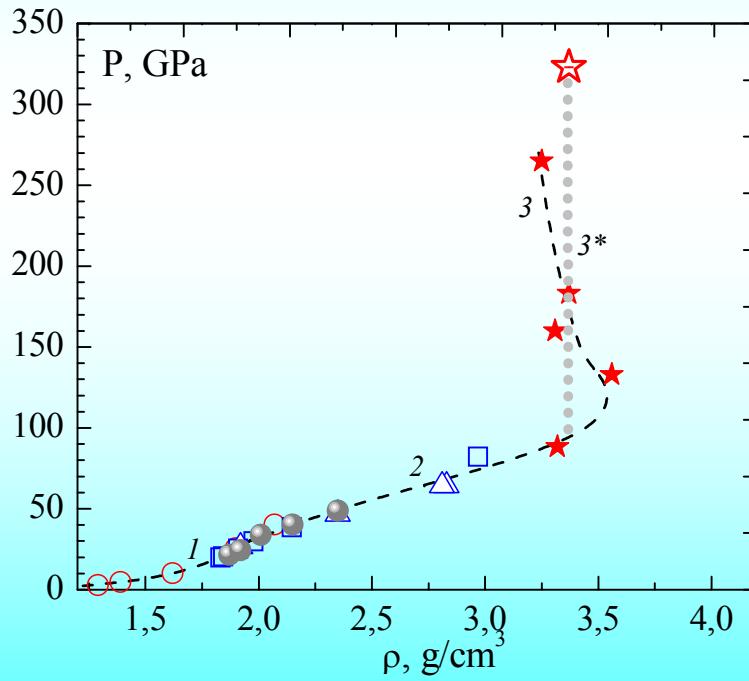
November 23-24, 2011



Sarov, 2011

Chernogolovka, 2010
EMMI, GSI, 2011

Thank you!



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and by MIPT Education Center "Physics of High Energy Density Matter" and by **Extreme Matter Institute (EMMI)**