PHASE TRANSITIONS AND BULK PROPERTIES OF THE TERNARY SYSTEM WATER-1-PROPANOL-n-HEXANE NEAR THE CRITICAL POINT

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The phase equilibrium of observed ternary system at low temperatures is investigated by authors [1–3] and at high temperatures by authors [4]. In this work by the method of free-ballast constant volume piezometer [5] the data on p, ρ, T relations of water-1-propanol-hexane system (0.3333 mol.fractions) is obtained in the wide diapason of state parameters.

Isochoric P-T break point technique was applied to accurately determine phase transition p_s, ρ_s, T_s and the critical point p_k, ρ_k, T_k parameters. The measured p, ρ, T properties in the one-phase region including phase transition curve and supercritical conditions were used to develop multiparametric polynomial type equation of state in the form of expansion of the pressure into series in powers of the reduced density $\omega = \rho/\rho_k$ and the reduced temperature $\tau = T/T_k$: $p = RT\rho_m \left[1 + \sum_{i=1}^m \sum_{j=0}^n a_{ij} \omega^i/\tau^i\right]$. Average relative deviation between the measured and calculated values of pressure is less than 1.3% in the whole measured temperature and pressure ranges. The results of the present work showed that volumetric behavior of the ternary system in the critical regions just like one-component system.

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