

ATOMIC DIFFUSION MECHANISMS IN TITANIUM CARBIDE

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Transition-metal carbides and nitrides combine ceramic-like properties (high melting point, hardness, and corrosion resistance) with metal-like properties (high electrical and thermal conductivity). These compounds are therefore highly suitable for many existing and emerging high-temperature applications [1]. Understanding atomic defects and diffusion in refractory compounds is important for controlling their properties during manufacturing and service [2]. While monovacancies can fully account for the self-diffusion in Group-IV metal nitrides [3], understanding the diffusion mechanisms of metal atoms in the corresponding carbides represents a challenge due to the extremely high formation energy of a metal vacancy [4,5]. An overview of recent *ab initio* studies of point defects, defect clusters, and diffusion mechanisms in titanium carbide will be given [4–7]. The focus will be made on the Ti self-diffusion mechanisms in TiC mediated by various defect complexes involving vacancies, self-interstitials, and/or oxygen impurity atoms.

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